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## Carbonate dynamics in a tropical coastal system in the South China Sea featuring upwelling, river plumes and submarine groundwater discharge

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Abstract This study examined carbonate dynamics in the northwestern South China Sea (NWSCS), an area jointly influenced by upwelling, river plumes and submarine groundwater discharge. Data were obtained from two cruises conducted in summer 2009 and 2012. In 2009, a high salinity-low temperature water mass occurred nearshore off northeastern Hainan Island, indicative of upwelling, commonly referred to as HNEU. A river plume fueled primarily by local rivers and characterized by low salinity and high temperature was observed in the NWSCS off the mainland roughly along the 30 m isobath. In 2012, coastal upwelling off northeastern Hainan Island was not detectable at the surface, but was observed at a different location off eastern Hainan Island (HEU). River plume waters in 2012 were patchily distributed, with a low salinity zone further westerly than that in 2009 and another on the mid-shelf of the NWSCS sourced from the Pearl River which reached out ~250 km from the mouth of the Pearl River Estuary. In 2009, elevated dissolved inorganic carbon (DIC) and total alkalinity (TA) occurred in the coastal plume, where submarine groundwater discharge contributed DIC and TA additions of  $38.9\pm20.5$  and  $42.5\pm22.3$  µmol kg<sup>-1</sup>, respectively, with a DIC/TA ratio of  $\sim 0.92$ , which made a minor contribution to the variation of seawater partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>), pH and the aragonite saturation state index ( $\Omega_{arag}$ ). Additionally, high surface phytoplankton production consumed DIC of  $10.0\pm10.4 \,\mu\text{mol kg}^{-1}$  but did not significantly affect TA, which dominated pCO<sub>2</sub> drawdown in the coastal plume water and increased the pH and  $\Omega_{arag}$  at surface. Submarine groundwater discharge was also observed in the region influenced by upwelling, but to a lesser degree than that impacted by coastal plume. Lower pH and  $\Omega_{arag}$  and higher pCO<sub>2</sub> values than in offshore waters were observed downstream of the upwelling system, attributable largely to organic matter remineralization with a DIC addition of 23.8±8.4 µmol kg<sup>-1</sup>. In 2012, submarine groundwater discharge was not detected but high phytoplankton production dominated carbonate dynamics in the coastal plume water with a net DIC consumption of 104.2 µmol kg<sup>-1</sup>, which markedly drew down sea surface  $pCO_2$  and increased pH and  $\Omega_{arag}$ . In the Pearl River Plume, the solubility-driven  $CO_2$  sink exceeded biological  $CO_2$ uptake, resulting in an additional decrease of pH and  $\Omega_{arag}$  and increase of seawater pCO<sub>2</sub>. Taken together, this study demonstrated complex spatial and year-to-year variability, and the controls of the carbonate system under the joint modulations of upwelling, river plumes and submarine groundwater discharge. A first order estimate that considered the rise of atmospheric  $CO_2$  and seawater temperature further suggested a high risk of ocean acidification in this coastal area by the end of this century, which could be amplified under the stresses of river plumes, submarine groundwater discharge and organic matter remineralization.

Keywords Carbonate system, Northwestern South China Sea, Coastal upwelling, Submarine groundwater discharge, Ocean acidification

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#### 1. Introduction

Upwelling is a ubiquitous coastal ocean feature (e.g., Friederich et al., 2002; Torres, 2003; Cao et al., 2011; Xue et al., 2016). Stimulated by upwelled nutrients, biological primary production in upwelling zones is typically high and sustains high fisheries production (e.g., Risien et al., 2004; Barth et al., 2007; Jing et al., 2011; Xue et al., 2016). How upwelling impacts the ocean carbonate system and air-sea carbon dioxide  $(CO_2)$  flux has been a subject of considerable interest (e.g., Feely et al., 2008; Huang et al., 2015; Xu et al., 2016), although it remains controversial. In many upwelling systems with a high upwelling intensity, such as the central California and northern Chile systems (Friederich et al., 2002; Torres, 2003), the increase in primary productivity stimulated by nutrients cannot compensate for the free CO<sub>2</sub> brought up from depth, resulting in its release from the ocean (Torres, 2003; Santana-Casiano et al., 2009; Cai, 2011; Cao et al., 2011; Xue et al., 2016). However, in some upwelling systems, the higher nutrient supply and subsequent biological consumption of dissolved inorganic carbon (DIC) lead to strong net CO<sub>2</sub> uptake. These systems include the northern California upwelling system off Oregon, the upwelling system off the Galician coast, the Benguela upwelling system, and the western Canada upwelling system (e.g., Borges and Frankignoulle, 2002; Hales et al., 2005; Santana-Casiano et al., 2009; Evans et al., 2012; Cao et al., 2014). The CO<sub>2</sub> flux pattern varies temporally over different evolution stages and/ or spatially in different upwelling subzones even within a single system (e.g., Hales et al., 2005; Cao et al., 2014). In the Oregon coastal upwelling, it is initiated by a strong CO<sub>2</sub> source due to outcropping of CO2-enriched subsurface water to the very nearshore zone over short timescales. This upwelled water is then transported seaward and southward and partial pressure of  $CO_2$  ( $pCO_2$ ) is drawn down by biological productivity to  $\sim 200 \,\mu atm$  (1 atm=1.01325×10<sup>5</sup> Pa), far below the atmospheric  $pCO_2$  value (Feely et al., 2008).

Upwelling not only brings up water with high CO<sub>2</sub> concentrations, but also lower pH and aragonite saturation state index ( $\Omega_{arag}$ ), which makes upwelling systems prone to ocean acidification (e.g., Feely et al., 2008, 2012; Lui et al., 2015; Xue et al., 2016). The dynamics of pH and  $\Omega_{arag}$ , however exhibit strong spatio-temporal variations and are subject to complex physical and biogeochemical controls within the water column (Cao et al., 2011; Xue et al., 2016). At the airsea interface,  $CO_2$  degassing (source) reduces seawater  $pCO_2$ but augments pH and  $\Omega_{arag}$ . In contrast, CO<sub>2</sub> uptake (sink) increases seawater  $pCO_2$  but reduces pH and  $\Omega_{arag}$  (Xu et al., 2016; Xue et al., 2017). Moreover, these effects are often closely associated with local water circulation and the supply of source materials. For instance, river plumes alter winddriven upwelling circulation by changing water column buoyancy and density structure (Gan et al., 2009; Dai et al., 2014), while submarine groundwater discharge contributes substantial material fluxes to the water column, which may enhance coastal acidification (e.g., Wang et al., 2014). Breakdown and quantification of the relative contributions from upwelling, river plumes and submarine groundwater discharge to carbonate dynamics in the coastal ocean are challenging and thus rarely reported (Gu et al., 2012; Luo et al., 2017; Dai et al., 2021).

The South China Sea is the largest marginal sea of the North Pacific Ocean (Dai et al., 2013 and references therein). On its northern shelf, two prominent coastal upwelling systems have been previously reviewed by Hu and Wang (2016). One is the Eastern Guangdong upwelling along the coast between the cities Shanwei and Shantou (e.g., Cao et al., 2011; Gan et al., 2009, 2010; Han et al., 2012). The other occurs off Hainan Island in the northwestern South China Sea (NWSCS) (e.g., Lü et al., 2008; Lin et al., 2016a, 2016b; Dong et al., 2017), which is modulated by the upwelling-favorable wind, flow-topography interaction, and Ekman pumping (e.g., Jing et al., 2011, 2015; Hu and Wang, 2016; Lin et al., 2016a, 2016b; Xie et al., 2017).

Our study area is located in the NWSCS featuring upwelling off Hainan Island, where coral reefs are widely distributed (Wang et al., 2014; Dong et al., 2017). Dong et al. (2017) examined the influence of coastal upwelling on a fringing reef system on the eastern coast of Hainan Island, and indicated that upwelled waters with low pH and  $\Omega_{arag}$ pose a high ocean acidification risk to this nearshore coral reef system. Wang et al. (2014) suggested that submarine groundwater discharge is another important contributor to acidification in a coastal coral reef system in Sanya Bay, located southeast of Hainan Island. However, there has been thus far no quantitative assessment of carbonate dynamics in the NWSCS that considers upwelling, river plumes and submarine groundwater discharge. Lack of this knowledge hampers our ability to assess the current status and predict future directions of the marine ecosystem, especially coral reefs that are of great societal concern.

This study investigates the carbonate system in the NWSCS using data collected during two summer cruises in 2009 and 2012. The changes in DIC and total alkalinity (TA) and subsequently the ocean acidification attributed to different water masses/processes were quantified, and the potential risks of ocean acidification to the marine ecosystem were predicted in the context of global climate change.

### 2. Materials and methods

### 2.1 Study area

Modulated primarily by the Asian monsoon, the NWSCS in summer is influenced by river plumes from the Pearl River and local rivers, as well as coastal upwelling (Hu and Wang, 10,1007/e11430-021-9963-8 2016; Lin et al., 2016a, 2016b; Dong et al., 2017). The Pearl River is the second largest river in China in terms of freshwater discharge with an annual runoff of  $3.26 \times 10^{11} \text{ m}^3$ , of which ~80% takes place during the wet season between April and October (Dai et al., 2014; Guo and Wong, 2015). The Pearl River Plume on the northern shelf of the South China Sea is a prominent feature in summer, which may extend southeastward up to a few hundred kilometers from the mouth of the estuary (e.g., Gan et al., 2009, 2010; Han et al., 2012; Yang et al., 2021; Zhao et al., 2021). Prior studies also suggest that the Pearl River Plume trajectory can be highly variable depending on wind stress and coastal currents. Coastal rivers, such as the Jianjiang River originating from the mainland, and the Wanquan and Wenchang Rivers from Hainan Island, may also influence the study area, but their discharges account for <2% of the Pearl River discharge (Lin et al., 2016b; Dong et al., 2017). Due to these very small discharges from coastal rivers, coastal plume waters were only observed nearshore (Lin et al., 2016b; Dong et al., 2017).

Coastal upwelling in the NWSCS typically starts in April, peaks from mid-July to mid-August, and disappears in September; in most cases it lasts 1–2 weeks (Hu and Wang, 2016). Upwelling in the NWSCS has been observed off the eastern coast of Hainan Island between the coasts of the cities of Wenchang and Wanning (so-called HEU), and off the northeastern coast of Hainan Island between the cities of Zhanjiang and Haikou (referred to as HNEU) (Lin et al., 2016b; Dong et al., 2017).

The submarine groundwater discharge is a particularly prominent feature at the highly permeable sediment-water interface, contributing abundant dissolved solutes to the water column (Gu et al., 2012; Liu et al., 2012, 2014; Luo et al., 2017; Tan et al., 2018). On the northern South China Sea shelf, surface sediments show a grain size gradation from gravel inshore to silt offshore, which consists of terrigenous and biogenous detritus, as well as authigenic minerals (Zhang et al., 2003; Liu et al., 2012). Results from radium isotopes (<sup>226</sup>Ra and <sup>228</sup>Ra) estimated the submarine groundwater discharge on the northern South China Sea shelf to be  $2.2 \times 10^8$  -  $3.7 \times 10^8$  m<sup>3</sup> d<sup>-1</sup>, equivalent to 12–21% of the Pearl River discharge. In turn, the DIC flux via submarine groundwater discharge was estimated to be  $153 \times 10^9$ - $347 \times 10^9$  mol yr<sup>-1</sup>, or equivalent to ~23-53% of the riverine DIC export flux (Liu et al., 2012). Recently, Dai et al. (2021) further confirmed the substantial contribution of submarine groundwater discharge to the DIC and TA in the NWSCS. In both studies, the DIC/TA ratio of the submarine groundwater discharge end-member varied in the range of  $\sim 0.9-1.1$ , which was however much lower than that in the Sanya Bay of ~3.3 (Wang et al., 2014). The different DIC/TA ratios of submarine groundwater discharge end-member suggested their spatial variability, and the need to better understanding of the influence of submarine groundwater discharge on the carbonate dynamics in the NWSCS.

### 2.2 Sampling and analysis

Sampling was conducted onboard the *R/V Dongfanghong II* in 2009 (Jul. 17–Aug. 16) and 2012 (Jul. 29–Aug. 21) in NWSCS (Figure 1). At each sampling station, temperature, salinity and depth were recorded with a Seabird® SBE 911 or 917 Conductivity-Temperature-Depth (CTD) sensor package. The distributions of temperature (T), salinity (S) and the *T-S* diagram during these two cruises were published in Lin et al. (2016a, 2016b) and Meng et al. (2017). During both cruises, water samples were collected at different depths throughout the water column with 30 L Go-Flo or 12 L Niskin bottles mounted on a rosette sampling assembly.

Samples for dissolved oxygen (DO), DIC and TA were taken with Tygon® tubing free of air bubbles, with ample sample overflow to minimize any contamination from atmospheric oxygen or CO<sub>2</sub>. Samples for DO were collected with 60 mL biological oxygen demand bottles and fixed with Winkler reagents (Carpenter, 1965). Both DIC and TA samples were poisoned with HgCl<sub>2</sub>-saturated solution upon sample collection; DIC samples were measured on shipboard within 24 h of sampling, and TA samples were stored in the dark and returned to a land-based laboratory for analysis within two months of collection. Samples for soluble reactive phosphorus (SRP or phosphate) measurements were filtered with 0.45  $\mu$ m polyacetate filters and stored in 120 mL high-density polyethylene bottles and analyzed onboard within 24 h.

The DIC was measured by collecting and quantifying the CO<sub>2</sub> released from the sample upon acidification with a nondispersive infrared detector (NDIR, Li-Cor<sup>®</sup> 7000), with a measurement precision better than  $\pm 0.1\%$ ; TA was determined by potentiometric Gran titration with a precision of  $\pm 2.0 \ \mu\text{mol kg}^{-1}$  (Cai et al., 2004). Measurements of DIC and TA were detailed in Cai et al. (2004). The accuracy of DIC and TA measurements was determined by calibration against certified reference materials provided by A. G. Dickson, Scripps Institution of Oceanography, U.S.A to an accuracy of better than  $\pm 2.0 \,\mu\text{mol kg}^{-1}$ . Seawater pH,  $\Omega_{arag}$  and influences of processes on  $pCO_2$  were calculated from measured DIC and TA using the CO2SYS program (Pierrot et al., 2006), in which the dissociation constants for carbonic acid were obtained from Mehrbach et al. (1973), as refitted by Dickson and Millero (1987), and the dissociation constant for  $HSO_4^-$  was determined as per Dickson (1990). The relationship of total boron with salinity followed Uppstrom (1967).

During the cruises, surface water was continuously pumped into analytical instruments from a side intake at a depth of  $\sim$ 5 m. Temperature and conductivity were measured 10.1007/c111420-021-0062-8



Figure 1 Maps showing the study area (a) and location of sampling stations during two summer cruises on July 17–Aug. 16, 2009 (b) and July 29–Aug. 21, 2012 (c) in the northwestern South China Sea (NWSCS). Also shown are the upwelling systems off the northeastern coast of Hainan Island (HNEU) in 2009 and off the eastern coast of Hainan Island (HEU) in 2012, following the description in Lin et al. (2016b). Vertical distributions of parameters measured along transects (T0 to T3) marked in red in panels (b) and (c) are detailed in the text. The Pearl River in southern China, the Jianjiang River in Leizhou Peninsula, and the Wanguan and Wenchang Rivers in Hainan Island are also shown.

continuously using an Idronaut Multiparameter "Flow Through" CTD recorder. A continuous-flow underway system with a cylinder-type equilibrator filled with plastic balls was used for air-sea equilibration (Zhai et al., 2005a, 2005b). The CO<sub>2</sub> mole fraction in dry air (xCO<sub>2</sub>) was measured continuously using a Li-Cor<sup>®</sup> 7000 NDIR detector, which was calibrated against a series of CO<sub>2</sub> gas standards. The overall uncertainty of the xCO<sub>2</sub> measurements and pCO<sub>2</sub> data processing was <1% (Zhai et al., 2005a, 2013). Air pCO<sub>2</sub> was determined every 4–12 h, and the bow intake from which atmospheric air was pumped was installed ~6–10 m above the water surface to avoid ship contamination. The pCO<sub>2</sub> data used in this study have been published in Li et al. (2020).

DO samples were measured spectrophotometrically at 466 nm onboard within 4 h of sampling (Labasque et al., 2004). Samples were placed in a constant temperature bath at

25.0±0.1°C for at least 1 h before measurement. Soluble reactive phosphorus was determined with the phosphomolybdenum blue method (Du et al., 2013) and run with an AA3 Auto-Analyzer (Bran-Lube, GmbH). The detection limit was 0.08  $\mu$ mol L<sup>-1</sup> and the analytical precision was better than ±2% (Han et al., 2012). The DO saturation degree (DO%) was the ratio of the observed DO to DO saturation, where the latter was calculated according to the empirical formula of Benson and Krause (1984).

# 2.3 Quantification of nutrient and carbonate dynamics with two or three end-member mixing models

A two or three end-member mixing model, as described in Dai et al. (2013), Cao et al. (2014) and Han et al. (2012), is used to predict their conservative concentrations and their comparisons with the measurements are used to quantify

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nutrient and carbonate dynamics mediated by different processes. In a three end-member mixing model, the mixing scheme was established with salinity and potential temperature ( $\theta$ ) as conservative tracers:

$$S_1 \times F_1 + S_2 \times F_2 + S_3 \times F_3 = S_x, \tag{1}$$

$$\theta_1 \times F_1 + \theta_2 \times F_2 + \theta_3 \times F_3 = \theta_x, \tag{2}$$

$$F_1 + F_2 + F_3 = 1, (3)$$

where  $S_x$  and  $\theta_x$  are the salinity and potential temperature of the water samples;  $S_1$ ,  $S_2$ ,  $S_3$  and  $\theta_1$ ,  $\theta_2$ ,  $\theta_3$  are the salinity and potential temperature of different end-members, respectively;  $F_1$ ,  $F_2$  and  $F_3$  are the respective fractional contributions by different end-members. Note that TA is also frequently used as a conservative tracer in three end-member mixing models while salinity is typically used in two endmember mixing models given its conservative nature.

The concentration of a chemical (X, representing the DIC or  $PO_4^{3-}$  concentration) from the conservative mixing of these three different end-members ( $X^{\text{conservative}}$ ) can be predicted as:  $K^{\text{conservative}} = K \times E + K \times E + K \times E$  (4)

$$X^{\text{conservative}} = X_1 \times F_1 + X_2 \times F_2 + X_3 \times F_3.$$
(4)

Finally, in a three end-member mixing model, the difference between field measurements ( $X^{\text{measurements}}$ ) and conservative values ( $X^{\text{conservative}}$ ) obtained (defined as  $\Delta X$ ), as:

$$\Delta X = X^{\text{measurements}} - X^{\text{conservative}},$$
(5)

where positive/negative values indicate non-conservative addition/removal of DIC and nutrients from other different sources/processes.

#### 3. Results

#### 3.1 Hydrographic setting

In the present study, the influence of coastal upwelling and plume water during both cruises was clearly reflected in the T-S diagrams and their spatial and cross-sectional distributions (Figures 2–7). In 2009, surface water with relatively low temperature (<28°C) and high salinity (>33) was observed at nearshore stations between the cities of Zhanjiang and Haikou (stations E500, QD02 and QD03) (Figure 2a, 2b), suggesting the influence of HNEU (Lin et al., 2016b). A similar T-S diagram for upwelled and offshore waters also confirmed the dominance of offshore subsurface water in the upwelling center (Figure 3a). In contrast, HNEU was not observed during the 2012 cruise, compounded by the fact that the area was covered by low salinity water (Lin et al., 2016a). However, upwelling evidenced by low sea surface temperature (SST, 25.4–27.0°C) and high sea surface salinity (SSS, >33) was observed at nearshore stations between the cities of Wenchang and Wanning (Figure 2d, 2e), indicating the appearance of HEU (Lin et al., 2016b).

In 2009, as reported by Yang et al. (2021), another water

mass characterized by high SST (>29°C) and low SSS (<32) was detected at stations QD01, F45 and E600 along the coast between the cities Maoming and Zhanjiang (Figure 2a, 2b), and was very likely influenced by the plume water from local rivers (i.e., by coastal plume water, defined as salinity <33.0). The influence of this low salinity plume water is also evident in the *T*-S diagram in Figure 3a. In 2012, however, surface water characterized by exceptionally low salinity (<32.0) was also observed at nearshore station QD00 (Figure 2e), a result comparable to that obtained in 2009. In addition, plume water from the Pearl River characterized by low SSS (<32.0) and high SST (>29.3°C) was also observed at off-shore stations along transect T0 (Figure 2d, 2e) (Dai et al., 2021), which was not observed in 2009.

# 3.2 Horizontal and vertical distributions of carbonate parameters

Horizontal and vertical distributions of temperature, salinity, DIC, TA and pH at 25°C (pH<sub>@25</sub>),  $\Omega_{arag}$  at 25°C ( $\Omega_{arag@25}$ ),  $pCO_2$  at 25°C ( $pCO_{2(a25)}$ ) and  $pCO_2$  at *in-situ* temperature  $(pCO_{2(in-situ)})$  in both cruises are shown in Figures 4–7. In 2009, SST ranged from 26.0 to 30.6°C, increasing with distance from shore (Figure 4a), and SSS varied 31.3 to 34.0 (Figure 4b). As mentioned earlier, high SST (>29.0°C) and low SSS (<33.0) at stations of QD01, F45 and E600 are indicative of coastal plume water; and low SST (<28°C) at stations E500, QD02 and QD03 is indicative of the influence of coastal upwelling. The range of DIC and TA was 1840-1940 and 2160–2230 µmol kg<sup>-1</sup>, respectively, with high values in the upwelling area (1910–1940  $\mu$ mol kg<sup>-1</sup> for DIC and 2210–2230  $\mu$ mol kg<sup>-1</sup> for TA), associated with their high salinity (Figure 4b-4d). Low DIC and TA values were observed in coastal plume water (1840–1860  $\mu$ mol kg<sup>-1</sup> for DIC and 2160–2190  $\mu$ mol kg<sup>-1</sup> for TA) (Figure 4c, 4d). Low  $pH_{@25}$  (8.03-8.08) and  $\Omega_{arag@25}$  (3.05-3.35) were also observed in the upwelling area, but high  $pH_{@25}$  (8.12–8.19) and  $\Omega_{arag@25}$  (3.37–3.74) were observed in coastal plume water (Figure 4e, 4f). Similar  $pCO_{2@25}$  and  $pCO_{2@in-situ}$  distribution patterns were obtained with high values in the upwelling center (360-420 µatm) and low values in coastal plume water (300–360 µatm) (Figure 4g, 4h). Offshore, surface distributions of temperature, salinity, DIC, TA, pH<sub>@25</sub>,  $\Omega_{arag@25}$ ,  $pCO_{2@25}$  and  $pCO_{2@in-situ}$  exhibited a relatively uniform distribution pattern with a range of 29.0-30.3°C, 32.6-33.8,  $1860-1890 \ \mu mol \ kg^{-1}$ ,  $2190-2220 \ \mu mol \ kg^{-1}$ , 8.10-8.15, 3.40-3.60, 280-340 µatm and 360-380 µatm, respectively (Figure 4a-4h).

Vertically, coastal plume water with low DIC and TA was observed in the upper mixed layer at stations E600 and QD01 along transects T1 and T2, associated with their low salinity (Figure 5). Additionally, high pH<sub>@25</sub> and  $\Omega_{arag@25}$ , and low  $pCO_{2@25}$ , and  $pCO_{2@in-situ}$  were observed in coastal plume 10.1007/s11430-021-9963-8



**Figure 2** Surface distribution of temperature, salinity and carbon dioxide partial pressure ( $pCO_2$ ) from measurements in 2009 ((a)–(c)) and 2012 ((d)–(f)). Red dots mark cities shown in Figure 1b. Crosses in (a) and (d) are sampling stations; HNEU and HEU are the same as in Figure 1.



**Figure 3** Relationship between potential temperature  $(\theta, ^{\circ}C)$  and salinity (*T-S* diagram) in summers of 2009 (a) and 2012 (b). Open symbols are observed data in the indicated areas. Red solid circles in panel (b) are the end-members (i.e., E1 (Offshore Surface Water), E2 (Offshore Subsurface Water) and E3 (Plume Water)) used in the three end-member mixing model. Red solid lines represent the hypothetical conservative mixing lines between different end-members.

water as previously reported in the plume center in other river estuaries, such as the Changjiang Estuary and the Pearl River Estuary (Cao et al., 2011; Chou et al., 2013). It is important to note that the influence of coastal plume water was limited to the surface. Beneath the plume water, the influence of coastal upwelling was conspicuous, as evidenced by the shoreward transport of deep water with high salinity (>34) and low temperature (<21°C) that outcropped nearshore (Figure 5). In the upwelling system, high DIC, TA,  $pCO_{2@25}$ , and  $pCO_{2@in-situ}$ , and low pH<sub>@25</sub>, and  $\Omega_{arag@25}$  water parcels upwelled from the offshore surface, resulting in outcrops of  $CO_2$ -enriched subsurface water nearshore, as previously reported (Dong et al., 2017).

In 2012, SST and SSS exhibited a range of 25.4–30.6°C and 29.0–34.0, respectively (Figure 6a, 6b). At HEU, surface water in the upwelling system was characterized by low SST (<27.0°C) and high SSS (>33.5). In contrast, surface water at QD00 was influenced by the coastal plume water, and at offshore stations along transect T0 was influenced by the Pearl River Plume characterized by high temperature (>29.0°C) and low salinity (<31.0), as noted in previous studies (Cao et al., 2011; Lin et al., 2016b; Dai et al., 2021;

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**Figure 4** Sea surface distribution of (a) temperature, SST, (b) salinity, SSS, (c) dissolved inorganic carbon, DIC, (d) total alkalinity, TA, (e)  $pH_{@25}$ , (f)  $\Omega_{arag@25}$ , (g)  $pCO_{2@jn-situ}$  in 2009. Salinity and temperature are surface data obtained at 1 m using CTD;  $pCO_{2@jn-situ}$  is obtained with a continuous-flow underway system;  $pCO_{2@25}$  represents  $pCO_2$  values normalized to 25°C, following method of Takahashi et al. (1993). Red dots mark cities shown in Figure 1b.

Yang et al., 2021). In coastal plume water, surface water was characterized by low DIC (<1780  $\mu$ mol kg<sup>-1</sup>), TA (<2100  $\mu$ mol kg<sup>-1</sup>),  $pCO_{2@25}$  (<280  $\mu$ atm) and  $pCO_{2@in-situ}$ 

(<280 µatm), and high  $pH_{@25}$  (>8.10) and  $\Omega_{arag@25}$  (>3.60) (Figure 6c–6h). Additionally, low DIC (1780–1840 µmol kg<sup>-1</sup>) and TA (2090–2150 µmol kg<sup>-1</sup>)



Figure 5 Vertical distributions of temperature, salinity, dissolved inorganic carbon (DIC), total alkalinity (TA),  $pH_{@25}$ ,  $\Omega_{arag@25}$ ,  $pCO_{2@25}$  and  $pCO_{2@in-situ}$  (definition of parameters as in Figure 4) along different transects (T1 to T3) in 2009. Salinity and temperature are obtained from 1 m using CTD. Upwelling Center and Coastal Plume Water are marked with red rectangles.

(Figure 6c, 6d) were also observed in the Pearl River Plume, while  $pH_{@25}$ ,  $\Omega_{arag@25}$ ,  $pCO_{2@25}$  and  $pCO_{2@in-situ}$  values were comparable to those of offshore surface water (Figure 6e–

d in the Pearl River Plume, 6h). Surface water characterized by high DIC  $pCO_{2@in-situ}$  values were (1870–1930 μmol kg<sup>-1</sup>), TA (2180–2220 μmol kg<sup>-1</sup>), surface water (Figure 6e–  $pCO_{2@25}$  (360–380 μatm) and  $pCO_{2@in-situ}$  (380–420 μatm), https://engine.scichina.com/doi/10.1007/s11430-021-9963-8



**Figure 6** Sea surface distributions of (a) temperature, SST, (b) salinity, SSS, (c) dissolved inorganic carbon, DIC, (d) total alkalinity, TA, (e)  $pH_{@25}$ , (f)  $\Omega_{arag@25}$ , (g)  $pCO_{2@25}$  and (h)  $pCO_{2@in-situ}$  (see text for definition of parameters) in 2012. Salinity and temperature are surface data obtained at 1 m using CTD.  $pCO_{2@in-situ}$  is obtained with a continuous-flow underway system.  $pCO_{2@25}$  are the  $pCO_2$  measurements normalized to 25°C following method of Takahashi et al. (1993). Red dots mark cities shown in Figure 1b.

and low pH<sub>@25</sub> (8.03–8.06) and  $\Omega_{arag@25}$  (3.10–3.30) values, was observed at HEU (Figure 6c–6h), suggesting the influence of CO<sub>2</sub>-enriched subsurface water. Offshore, tempera-

ture, salinity, DIC, TA,  $pH_{@25}$ ,  $\Omega_{arag@25}$ ,  $pCO_{2@25}$  and  $pCO_{2@in-situ}$  in surface water varied in the range 27.8–30.2°C, 32.8–33.8, 1850–1880 µmol kg<sup>-1</sup>, 2180–2215 µmol kg<sup>-1</sup>,

https://engine.scichina.com/doi/10.1007/s11430-021-9963-8



Figure 7 Vertical distributions of temperature, salinity, dissolved inorganic carbon (DIC), total alkalinity (TA),  $pH_{@25}$ ,  $\Omega_{arag@25}$ ,  $pCO_{2@25}$  and  $pCO_{2@in-situ}$  (parameters defined as in Figure 4) along different transects (T0 to T3) in 2012. Salinity and temperature were obtained at 1 m using CTD. Upwelling Center, Coastal Plume Water and Pearl River Plume are marked with red rectangles.

8.08–8.12, 3.40–3.55, 300–350 µatm and 380–410 µatm, respectively, with a rather uniform distribution pattern

μatm and 380–410 μatm, (Figure 6a–6h). Vertically, the influence of plume water resulted in low DIC, TA,  $pCO_{2@25}$  and  $pCO_{2@in-situ}$ , and high https://engine.scichina.com/doi/10.1007/s11430-021-9963-8 pH<sub>@25</sub> and  $\Omega_{arag@25}$  values in the upper mixed layer at offshore stations along transect T0 and at station QD00 (Figure 7). At nearshore stations along transect T3, the uplifted isopleths of DIC, TA, pH<sub>@25</sub>,  $\Omega_{arag@25}$ ,  $pCO_{2@25}$  and  $pCO_{2@in-situ}$  suggested upwelling of CO<sub>2</sub>-enriched subsurface water from offshore (Figure 7).

#### 4. Discussion

#### 4.1 Variability of river plumes in the NWSCS

The occurrence of river plumes is frequently reported in coastal areas in the NWSCS (Lin et al., 2016b; Dong et al., 2017; Yang et al., 2021). In 2009, the coastal plume water TA-S relationship was described by the linear equation v=36.7x+1088 (coefficient of determination,  $R^2=0.81$ ). Prior studies suggested that TA concentrations from the freshwater end-member in the mainland (such as the Jianjiang River) and from Hainan Island (e.g., the Wanquan and Wenchang Rivers) varied over a narrow range of 460-737 µmol kg<sup>-</sup> (Dai et al., unpublished data; Dong et al., 2017). The observed TA intercept was much higher than that of the freshwater end-member of coastal local rivers, indicating contributions of other TA sources. In coastal plume water at surface, the activities of  $^{226}$ Ra and  $^{228}$ Ra (in dpm 100 L<sup>-1</sup>) ranged from 8.5 to 12.5 and from 20.0 to 30.0, respectively. which was much higher than those in offshore surface water (Wang et al., unpublished data). Relatively high <sup>228</sup>Ra/<sup>226</sup>Ra ratios. 2.93 to 3.59, were also observed in coastal plume water. Among the potential sources of radium to this region, the average <sup>228</sup>Ra/<sup>226</sup>Ra ratio of coastal local rivers was 2.52  $\pm 0.02$ , and that of coastal groundwater was 7.68 $\pm 2.24$  (Dai et al., 2021). Thus, such high <sup>228</sup>Ra/<sup>226</sup>Ra ratios in coastal plume water must have resulted from submarine groundwater discharge. In this region (e.g., station QD01), the influence of submarine groundwater discharge can be discerned in the whole water column from the relatively high <sup>228</sup>Ra/<sup>226</sup>Ra ratios (Wang et al., unpublished data).

In 2012, two low salinity water parcels were observed at station QD00 and at offshore stations along transect T0. At station QD00, a linear regression provided a good fit to the water column relationship between TA and salinity (y=44.4x +700,  $R^2=0.94$ ). The intercept was consistent with the freshwater end-member of coastal local rivers (Dai et al., unpublished data; Dong et al., 2017), thus confirming the influence of coastal plume water in 2012 as suggested by Lin et al. (2016b). For the upper water column along transect T0, a very good linear relationship was found between TA and S (y=33.5x+1232,  $R^2=0.90$ ), the intercept of which is consistent with the Pearl River freshwater end-member that ranged seasonally from 1190 to 2114 µmol kg<sup>-1</sup> (Guo et al., 2008). This confirms the influence of the Pearl River Plume as suggested by Dai et al. (2021).

To clarify the general effect of freshwater from coastal rivers, Lin et al. (2016b) computed the precipitation rate during these two summer cruises. Their results suggested that, in 2009, high precipitation rate in the Leizhou Peninsula brings considerable freshwater to nearshore areas off the northeastern coast of Hainan Island, resulting in a low-salinity, nearshore surface water mass, whereas the discharge from coastal rivers was much lower in 2012. Thus, the plume area influenced by coastal plume water was larger in 2009 than in 2012.

To examine the influence of the Pearl River Plume during both cruises, surface distributions of temperature and salinity during the summers of 2009 and 2012 are shown for the entire northern shelf in Appendix Figure S1 (https://link. springer.com). Results showed that the Pearl River Plume was observed on the northeastern shelf in 2009, and on the northwestern shelf in 2012. On the northern shelf, the orientation of the Pearl River Plume is suggested to be modulated mainly by the wind field (Dong et al., 2004; Ou et al., 2009; Xu et al., 2019). When southeasterly and easterly winds prevail, the Pearl River Plume generally extends along the western coast under the influence of the wind-driven westward coastal current and shoreward Ekman transport, and thus plume water is observed on the northwestern part of the shelf. In contrast, when southerly and southwesterly winds prevail, the Pearl River Plume typically expands in an eastward direction, which is often observed on the northeastern part of the shelf (Dong et al., 2004; Gan et al., 2009; Ou et al., 2009; Xu et al., 2019). In 2009, the Pearl River Plume was observed on the shelf on July 20. Subsequently, forced by southerly winds during July 20-25, surface plume water was transported eastwardly, and the plume center was observed on the northeastern shelf on July 25 (Appendix Figure S2b, S2c). The movement of plume water was also tracked by the currents (Appendix Figure S2b, S2c). In 2012, the Pearl River Plume characterized by high SST was observed in the Pearl River Estuary on August 3 (Appendix Figure S2f), and subsequently it was transported southward under the prevailing northwesterly wind stress, and plume water was finally observed on the northwestern offshore shelf on August 8 (Appendix Figure S2g).

It is interesting to note that the plume area influenced by the Pearl River was larger in 2012 than in 2009 during the cruises (Appendix Figure S1), whilst the freshwater discharge rate in Xijiang River (the largest river of the Pearl River system) for the period from July 10 to August 10 before and during sampling in 2012 was lower, ~69% of the 2009 average discharge (for the period from June 28 to July 28 before and during sampling). This suggests that the plume area is not only determined by the river discharge rate but also influenced by the wind filed (e.g., Chen et al., 2017; Fong and Geyer, 2001). Zu et al. (2014) also reported a larger plume area with a smaller river discharge rate on the northeastern shelf, which was mediated by both wind stress and river discharge. We contend that the tendency of the surface plume range to increase due to wind mixing far exceeded the tendency for it to shrink caused by reduced river discharge (Zu et al., 2014).

# 4.2 Processes controlling sea surface *p*CO<sub>2</sub> dynamics in the NWSCS

As shown in Figure 2, the distributions of sea surface  $pCO_2$  ( $pCO_{2@in-situ}$ ) showed high spatial variability. During both cruises, the highest  $pCO_2$  values (>400 µatm) were observed in the upwelling center, whereas the lowest values (<350 µatm) were observed in the plume water. Offshore, the sea surface  $pCO_2$  range was 360–410 µatm with a fairly uniform distribution pattern (Figure 2c, 2f). To identify the effects of temperature and water mass mixing on the dynamics of sea surface  $pCO_2$  and SST ( $lnpCO_2$ -SST) and between temperature-normalized sea surface  $pCO_2$  ( $npCO_2$ ) and SSS ( $npCO_2$ -SSS) were plotted in Figure 8. An annual average SST of 26°C was used for temperature normalization (Zhai et al., 2005a, 2013), following method of Takahashi et al. (1993).

The lower and upper limit of the annual average sea surface  $pCO_2$  on the northern South China Sea shelf, 350 and 390 µatm (Zhai et al., 2005a, 2013), respectively, are the intercepts of the linear regression equations indicated by the black dashed lines in Figure 8a, 8c. If the observed  $pCO_2$ varies between these two dashed lines, sea surface  $pCO_2$  is thus mainly controlled by temperature. In turn, for  $pCO_2$  values higher/lower than the upper/lower dashed line, other effects, may override temperature effects in regulating sea surface  $pCO_2$ . During both cruises, most  $pCO_2$  measurements from the upwelling and offshore areas occurred above or below these two dashed lines, leading to the conclusion that non-temperature effects were the main contributors to the spatial dynamics of sea surface  $pCO_2$ .

In Figure 8b, 8d,  $npCO_2$  in the upwelling area during both cruises is much higher than in the other sub-regions, supporting the dominance of upward transport of CO<sub>2</sub>-enriched subsurface water in upwelling systems (Cao et al., 2011; Xue et al., 2016). In the Pearl River Plume (Figure 10d) and coastal plume water during both cruises (Figure 8b, 8d), the constrained  $npCO_2$ -SSS relationships suggest that the mixing of plume water was characterized by low SSS and  $npCO_2$ , while offshore surface water exhibited high SSS and  $npCO_2$ . Thus, water mass mixing may dominate the spatial dynamics of sea surface  $pCO_2$  in these sub-regions. In contrast, offshore  $npCO_2$  is nearly constant during both cruises, showing that water mass mixing plays a minor role in regulating sea surface  $pCO_2$ .

Finally, phytoplankton production may also contribute to the spatial dynamics of sea surface  $pCO_2$ . In coastal plume water and the Pearl River Plume, supersaturated DO (DO% >110%) was determined in surface water (Figure 9e, 9f), indicating relatively high phytoplankton production in these sub-regions, which may also contribute to the low  $pCO_2$ . To evaluate the dynamics of sea surface  $pCO_2$  attributed to



**Figure 8** Relationships of  $\ln pCO_2$  vs. sea surface temperature, SST ( $\ln pCO_2$ -SST, left panels) and  $npCO_2$  vs. sea surface salinity, SSS ( $npCO_2$ -SSS, right panels) for surface waters in 2009 ((a), (b)) and 2012 ((c), (d)). In panels (a) and (c), red dashed lines indicate the logarithm of air  $pCO_2$ , and the two black dashed lines encompass an area where temperature dominates sea surface  $pCO_2$  variability (see text for details). In panels (b) and (d), red dots represent the end-members of Offshore Surface Water, Plume Water, River Water and Nearshore Surface Water as shown in Table 1; green and red solid lines represent the hypothetical conservative mixing lines between different end-members. In panels (c) and (d), red circles are the measurements in the Pearl River Plume.



**Figure 9** TA-Salinity, S ((a), (b)), DIC-S ((c), (d)), DO%-S ((e), (f)),  $pH_{@25}$ -S ((g), (h)),  $\Omega_{arag@25}$ -S ((i), (j)) and  $pCO_{2@25}$ -S ((k), (l)) relationships in summer 2009 (left panels) and 2012 (right panels) in various indicated locations. In panels (a)–(d), red dots represent the end-members of River Water (E1), Nearshore Surface Water (E2), Nearshore Subsurface Water (E3), and Submarine Groundwater Discharge (E4) as shown in Table 1; red solid lines represent the hypothetical conservative mixing lines between different end-members. Also shown in panel (a) are the three end-member mixing model of River Water, Nearshore Surface Water and Submarine Groundwater Discharge in Coastal Plume Water (mixing model 1, MM1) and the mixing of Nearshore Surface Water, Nearshore Subsurface Water and Submarine Groundwater Discharge in upwelling waters (mixing model 2, MM2).

phytoplankton production, hypothetical relationships of  $npCO_2$  vs. SSS in Offshore Surface Water, Plume Water, River Water and Nearshore Surface Water were also plotted (Figure 8b, 8d); their end-member values are summarized in Table 1. Results of this analysis indicated that in coastal plume water during both cruises, the  $npCO_2$  values from field measurements were much lower than along the conservative mixing line, suggesting  $pCO_2$  drawdown attributed to high phytoplankton production. In the Pearl River Plume, lower  $npCO_2$  values than along the conservative mixing line were also observed, but the  $pCO_2$  drawdown was much weaker than in coastal plume water.

In Figure 8a, 8c, if  $pCO_2$  measurements are higher (lower) than the air  $pCO_2$ , the area acts as a  $CO_2$  source (sink), resulting in a decrease (increase) of DIC and sea surface  $pCO_2$ . Thus,  $CO_2$  outgassing will result in drawdown of sea surface  $pCO_2$  in both upwelling and offshore areas, while  $CO_2$  intrusion will raise sea surface  $pCO_2$  in the Pearl River Plume and coastal plume water.

### 4.3 Dynamics of the carbonate system mediated by river plumes, biological processes and submarine groundwater discharge in the NWSCS

4.3.1 Dynamics of the carbonate system in River plumes As shown in section 4.1, the coastal plume water in 2009 was influenced by the submarine groundwater discharge. High DO%, pH<sub>@25</sub> and  $\Omega_{arag@25}$ , and low pCO<sub>2@25</sub> were also determined in this low salinity plume water (Figure 9e, 9g, 9i, 9k), indicative of high phytoplankton production. At the surface, coastal plume water seemed to be controlled by River Water, Nearshore Surface Water and Submarine Groundwater Discharge water masses, and the dynamics of

carbonate parameters were thus quantified using a three endmember mixing model using salinity and TA as tracers (see MM1 in Figure 9a). Using end-members summarized in Table 1, the contribution of submarine groundwater discharge ranged from 0.5% to 1.5% in surface water, with an average of  $0.9\pm0.5\%$ . Wang et al. (unpublished data) also quantified the fractions of submarine groundwater discharge during the same cruise using a three end-member mixing model with tracers of long-lived radium isotopes (<sup>226</sup>Ra and <sup>228</sup>Ra). Their result shows that the proportion of groundwater ranged from 0.1% to 1.4% in the coastal plume water with an average of 0.8±0.4%, agreed rather well with our calculations. Based on the high DIC and TA concentrations in the submarine groundwater discharge end-member, the DIC and TA additions from submarine groundwater discharge were estimated at  $38.9\pm20.5 \ \mu\text{mol kg}^{-1}$  (range=18.5–59.4  $\mu\text{mol kg}^{-1}$ ) and  $42.5\pm22.3 \ \mu\text{mol kg}^{-1}$  (range=20.1–64.8 (range=18.5-59.4 µmol kg<sup>-1</sup>), respectively. Finally, DIC removal of  $10.0\pm10.4$  µmol kg<sup>-1</sup> is estimated to balance the DIC budget, which was probably attributable to the high phytoplankton DIC uptake in plume water.

To further quantify the dynamics of pH,  $\Omega_{arag}$  and  $pCO_2$ mediated by submarine groundwater discharge and the phytoplankton production, the following predictions based on a three end-member mixing model were made below. Firstly, we calculated the conservative DIC following method described in Section 2.3 (defined as DIC<sup>conservative</sup>), and pH,  $\Omega_{arag}$ and  $pCO_2$  calculated from DIC<sup>conservative</sup> and TA<sup>measurements</sup> at 25°C were obtained with the CO2SYS program (defined as pH<sup>conservative</sup>,  $\Omega_{arag}$ <sup>conservative</sup> and  $pCO_2$ <sup>conservative</sup>, respectively). Secondly, DIC and TA additions from submarine groundwater discharge (SGD) were subtracted from DIC<sup>conservative</sup> and TA<sup>measurements</sup> (defined as DIC<sup>conservative-SGD</sup> and

Table 1Summary of end-member values of the properties of water masses in sub-regions influenced by Coastal Plume Water and coastal upwelling in2009, and by Coastal Plume Water and the Pearl River Plume in  $2012^{a^1}$ 

Cruise	End-members	Potential temperature (°C)	Salinity	DIC (µmol kg <sup>-1</sup> )	TA ( $\mu$ mol kg <sup>-1</sup> )
2009	River water		0	522.5	500.7
	Nearshore subsurface water	21.8	34.3	2008.9	2262.7
	Nearshore surface water	28.8	33.2	1879.7	2207.0
	Submarine groundwater discharge	22.5	24.3	4054.9	4420.4
2012	River water		0	522.5	500.7
	Plume water	29.3	30.3	1815.0	2105.5
	Offshore surface water	29.9	33.3	1888.1	2191.4
	Offshore subsurface water	20.6	34.4	2031.0	2252.0
	Nearshore surface water	29.3	32.6	1896.0	2196.0

a) The River Water end-member was obtained from Dai et al. (unpublished data) and Dong et al. (2017), while Plume Water is the average of low salinity surface water ( $\sim$ 5 m) in the plume center. Value of the Offshore Surface Water end-member is the average of offshore data at 5 m depth, and the Offshore Subsurface Water end-member is the average of data at  $\sim$ 50–70 m depth (see text). Nearshore Surface Water value correspond to those of surface water ( $\sim$ 5 m) at stations D102 in 2009 and QD21a in 2012; Nearshore Subsurface Water value correspond to those of subsurface water ( $\sim$ 35 m) at station E601. The Submarine Groundwater Discharge end-member was obtained from Wang et al. (unpublished data). DIC=dissolved inorganic carbon, TA=total alkalinity.

TA<sup>measurements-SGD</sup>), and pH,  $\Omega_{arag}$  and  $pCO_2$ subsequently calculated from DIC<sup>conservative-SGD</sup> were and TA<sup>measurements-SGD</sup> at 25°C (defined as pH<sup>conservative-SGD</sup>  $\Omega_{arag}^{conservative-SGD}$  and  $pCO_2^{conservative-SGD}$ , respectively). Their differences are the deviations attributed to submarine groundwater discharge. Thirdly, the difference in pH,  $\Omega_{arag}$ and  $pCO_2$  between the field measurements at 25°C (defined pH<sup>measurements</sup>  $\Omega_{arag}^{measurements}$ and  $pCO_2^{\text{measurements}}$ as respectively) and the conservative values (pH<sup>conservative</sup>,  $\Omega_{\text{arag}}^{\text{conservative}}$  and  $p \text{CO}_2^{\text{conservative}}$ ) were obtained, which are attributed to phytoplankton production. As a result, predicted changes in pH,  $\Omega_{arag}$  and pCO<sub>2</sub> attributed to submarine groundwater discharge were relatively minor (<0.01 for pH, <0.10 for  $\Omega_{arag}$  and <10 µatm for pCO<sub>2</sub>), and DIC and TA additions followed a ratio of ~0.92, while the high phytoplankton production in coastal plume water would reduce the  $pCO_2$  by 14.3±14.2 µatm (range=1.3-29.0 µatm) and increase the pH and  $\Omega_{arag}$  by 0.02±0.02 (range=0–0.04) and 0.11±0.11 (range=0.01-0.22), respectively.

In 2012, the influence of coastal plume water was also observed at station QD00. A two end-member mixing model between River Water and Nearshore Surface Water was established (Figure 9b, 9d). In surface water, the observed TA was almost exclusively on the mixing line, while observed DIC was much lower than the mixing line, suggesting DIC consumption via phytoplankton production. Associated with this DIC consumption, high DO%,  $pH_{@25}$  and  $\Omega_{arag@25}$ , and low  $pCO_{2@25}$  were indeed consistently observed (Figure 9f, 9h, 9j, 9l). In the coastal plume water, DIC removal via strong phytoplankton production was estimated at 104.2  $\mu$ mol kg<sup>-1</sup> at the surface, which drew down the sea surface  $pCO_2$  by 111.1 µatm but augmented pH and  $\Omega_{arag}$  by 0.13 and 0.73, respectively. It is noteworthy that in coastal plume water the influence of submarine groundwater discharge was only observed in 2009, suggesting temporal variability of the submarine groundwater discharge with varying precipitation rates that is in agreement with previous reports (Liu et al., 2012; Wang et al., 2014; Tan et al., 2018).

In the Pearl River Plume during 2012, the dynamics of TA, DIC and other carbonate parameters was likely controlled by mixing of different water masses and subsequently by the high phytoplankton production supported by nutrients from different end-members that showed high DO%, pH<sub>@25</sub> and  $\Omega_{arag@25}$ , and low  $pCO_{2@25}$  (Figure 9f, 9h, 9j, 9l). Three end-members were identified from the *T-S* relationship: Plume Water, Offshore Surface Water and Offshore Subsurface Water (Figure 3b). Below we quantify DIC dynamics using a three end-member mixing model with potential temperature and salinity as tracers as described in section 2.3. Since airwater CO<sub>2</sub> exchange has no effect on TA, we validated the model with TA. The resulting model-derived values are in good agreement with field observations with uncertainties averaging  $3.2\pm5.3 \,\mu\text{mol}\,\text{kg}^{-1}$ , thus strongly supporting our

model predictions. In surface water, the non-conservative portion of DIC ( $\Delta$ DIC) and nutrients ( $\Delta$ PO<sub>4</sub>) during mixing of different water masses were obtained. Total  $\Delta$ DIC is 18.7±20.2 µmol kg<sup>-1</sup>, a net result of biological activity and air-sea CO<sub>2</sub> exchange. Biological uptake of PO<sub>4</sub><sup>3-</sup> is 0.1±0.1 µmol kg<sup>-1</sup>, and thus DIC photosynthetic consumption (calculated as 106 $\Delta$ PO<sub>4</sub> based on the classic Redfield ratio (Redfield et al., 1963) is 7.9±6.0 µmol kg<sup>-1</sup>. To balance the DIC budget, DIC addition induced by air-sea CO<sub>2</sub> exchange is estimated as 26.5±19.5 µmol kg<sup>-1</sup>.

In the Pearl River Plume, average (±SD) sea surface  $pCO_2$ , pH and  $\Omega_{arag}$  due to conservative mixing, calculated from DIC<sup>conservative</sup> and TA<sup>conservative</sup> at the *in-situ* temperature using CO2SYS software, were 379.5±14.2 µatm, 8.05±0.01 and 3.43±0.05, respectively. Then, biological uptake supported by the nutrients in different end-members would draw down the  $pCO_2$  by 4.3±4.2 µatm and increase the pH and  $\Omega_{arag}$  by 0.01±0.01 and 0.02±0.02, respectively. Finally, CO<sub>2</sub> uptake in plume water would level up sea surface  $pCO_2$  by 10.4 ±15.3 µatm and draw down pH and  $\Omega_{arag}$  by 0.01±0.01 and 0.06±0.09, respectively, based on DIC mass balance.

#### 4.3.2 Dynamics of the carbonate system in upwelling

Rather different from the upwelled waters in 2012 where carbonate dynamics was dominated by the shoreward transport of offshore water without the influence of other water masses, the upwelled waters in 2009 was characterized by the influence of a water mass of high DIC, TA and  $pCO_{2@25}$ , and low DO%,  $pH_{@25}$  and  $\Omega_{arag@25}$  (Figure 9a, 9c, 9e, 9g, 9i, 9k). In 2009, similar to observations in coastal plume water at stations QD01, F45 and E600, surface water in the upwelling center (stations E500, QD02 and QD03) was characterized by higher <sup>226</sup>Ra and <sup>228</sup>Ra activities than offshore surface water (Wang et al., unpublished data). The observed higher <sup>228</sup>Ra/<sup>226</sup>Ra ratios (~2.85) than the endmember of coastal local rivers (2.52±0.02) suggested the influence of submarine groundwater discharge at the surface (Dai et al., 2021). Vertically, in the upwelling center, surface and subsurface <sup>228</sup>Ra activities and <sup>228</sup>Ra/<sup>226</sup>Ra ratios were higher than in bottom water (e.g., station E500, Wang et al., unpublished data), suggesting that submarine groundwater discharge additions occurred at the surface and subsurface transported from inshore waters. Thus, the influence of submarine groundwater discharge on DIC and TA dynamics could be estimated using a three end-member mixing model with salinity and TA as tracers, and a mixing scheme between Nearshore Surface Water, Nearshore Subsurface Water and submarine groundwater discharge (as shown in Figure 9a, see MM2). Using the end-members summarized in Table 1, the contribution of submarine groundwater discharge ranged from 0.1% to 0.9%, and averaged  $0.5\pm0.3\%$ , agreed rather well with results from a three end-member mixing model with tracers of <sup>226</sup>Ra and <sup>228</sup>Ra (ranged from

0.3% to 1.1% with an average of 0.6±0.3%) (Wang et al., unpublished data). Mean DIC and TA additions from submarine groundwater discharge were estimated at 22.3±12.6  $\mu$ mol kg<sup>-1</sup> (range=4.7–37.9  $\mu$ mol kg<sup>-1</sup>) and 24.3±13.8  $\mu$ mol kg<sup>-1</sup> (range=5.1–41.3  $\mu$ mol kg<sup>-1</sup>), respectively. Finally, a DIC addition of 23.8±8.4  $\mu$ mol kg<sup>-1</sup> (range =11.8–33.6  $\mu$ mol kg<sup>-1</sup>) was required to balance the DIC budget, was an addition probably attributable to DIC regeneration resulting from organic matter remineralization (Sunda and Cai, 2012; Su et al., 2017).

Additionally, the dynamics of pH,  $\Omega_{arag}$  and  $pCO_2$  mediated by submarine groundwater discharge and organic matter remineralization were quantified. The additions of submarine groundwater discharge made a minor contribution to the variability of pH,  $\Omega_{arag}$  and  $pCO_2$  (<0.01 for pH, <0.10 for  $\Omega_{arag}$  and <10 µatm for  $pCO_2$ ), a result consistent with that in coastal plume water in 2009. Thus, the dynamics of pH,  $\Omega_{arag}$ and  $pCO_2$  in upwelling waters were mostly attributed to organic matter remineralization, which would draw down the pH and  $\Omega_{arag}$  by 0.04±0.04 (range=0.02–0.06) and 0.24±0.09 (range=0.12–0.35), respectively, but increase the seawater  $pCO_2$  by 63.4±32.2 µatm (range=22.5–103.3 µatm).

# 4.4 Future scenarios of seawater acidification resulting from multiple drivers

As shown in section 4.3, the dynamics of the carbonate system in the NWSCS was mainly controlled by plume water from coastal local rivers, submarine groundwater discharge and organic matter remineralization. Additionally, increasing atmospheric CO<sub>2</sub> may facilitate its dissolution in seawater (Chou et al., 2013; Wang et al., 2014). However, the increasing seawater temperature may level up the seawater  $pCO_2$ , and partially cancel off the CO<sub>2</sub> sink (Chou et al., 2013; McNeil and Matear, 2007). In the present study, we predict pH and  $\Omega_{arag}$  in the NWSCS under the interaction of increasing atmospheric CO<sub>2</sub> and seawater temperature, river plume effects, submarine groundwater discharge influence and organic matter remineralization in 2100 with the following assumptions.

Firstly, due to the fact that the carbonate system for the entire water column at nearshore stations (<100 m depth) without the influence of coastal plume water and submarine groundwater discharge in 2009 is consistent with 2012, the nearshore carbonate system in 2009 is taken as the reference condition. Secondly, we assume that seawater  $pCO_2$  is in equilibrium with atmospheric CO<sub>2</sub>, and air CO<sub>2</sub> concentration would be 800 ppm (1 ppm=1×10<sup>-6</sup>) in 2100 (Chou et al., 2013; Wang et al., 2014). In addition, the sea surface temperature increases is assumed to be 2°C (Chou et al., 2013; McNeil and Matear, 2007). Thirdly, we assume that DIC and TA additions from submarine groundwater discharge remain unchanged, which equals ~40 µmol kg<sup>-1</sup>. Next, we assume

strong biological respiration in submarine groundwater discharge, which would raise the DIC concentration by ~24 µmol kg<sup>-1</sup>. Finally, we assume strong phytoplankton production in surface plume water, which will likely generate more organic matter and its sinking to bottom water for further decomposition (Sunda and Cai, 2012; Chou et al., 2013; Su et al., 2017). Given the DIC consumption of ~104.2 µmol kg<sup>-1</sup> in the surface water of the coastal plume water and 60% of the marine sources organic carbon settled to the subsurface and bottom water (Wu et al., 2017), remineralization of this organic carbon would increase DIC concentrations by ~60 µmol kg<sup>-1</sup> in bottom waters.

Our prediction showed that in 2100, the CO<sub>2</sub> sequestration regulated by both the increasing air CO<sub>2</sub> and seawater temperature would level up the DIC concentration by 122.1±20.0 µmol kg<sup>-1</sup>, and pH and  $\Omega_{arag}$  would be down to 7.78±0.01 and 2.12±0.22, respectively. Furthermore, the influence of submarine groundwater discharge and organic matter remineralization could reduce pH and  $\Omega_{arag}$  to 7.56 ±0.01 (range=7.54–7.58) and 1.39±0.18 (range=1.00–1.65). The predicted  $\Omega_{arag}$  values in bottom waters at some stations could be ~1.00, suggesting a high risk of seawater acidification in this coastal ecosystem at the end of this century.

In this coastal area, freshwater discharge from rivers is characterized by low DIC and TA, and therefore low carbonate ion concentrations and low  $\Omega_{arag}$ . Thus, the dilution of freshwater in the plume area may result in additional acidification (Chou et al., 2013; Zhai et al., 2015). Furthermore, the anticipated increase of nutrient and organic carbon loading would inevitably enhance the coastal water acidification (Wang et al., 2014). In the context of global climate change, all these processes/factors would make ocean acidification in coastal areas a much more dynamic and severe phenomenon, whose controlling mechanisms require future consideration.

#### 5. Conclusions

The present study reports for the first time the joint influence of coastal upwelling, submarine groundwater discharge and river plumes on carbonate dynamics in the NWSCS. While the coastal plume water featuring low seawater  $pCO_2$ , and high pH and  $\Omega_{arag}$  was evident during both cruises, the present study revealed the year-to-year dynamic of the Pearl River Plume associated with the variability of river discharge and coastal currents. Upwelling was observed during both cruises but at different locations under different wind stress, coastal currents, and the influence of freshwater. The influence of submarine groundwater discharge also showed yearto-year difference, regulated by the varying precipitation rates. Coastal acidification is a complex phenomenon driven by multiple physical and biogeochemical processes. Our first order predictions suggested that the increasing atmospheric  $CO_2$  and seawater temperature would pose a high risk of seawater acidification in this coastal ecosystem at the end of this century, which could be amplified under the stress of river plume effects, and influence of submarine groundwater discharge and organic matter remineralization.

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