

# Diurnal to interannual variability of sea surface $p\text{CO}_2$ and its controls in a turbid tidal-driven nearshore system in the vicinity of the East China Sea based on buoy observations

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## ABSTRACT

We examined the diurnal to seasonal dynamics of the sea surface partial pressure of carbon dioxide ( $p\text{CO}_2$ ) in a subtropical nearshore estuarine system, Hangzhou Bay, adjacent to the Changjiang Estuary in the vicinity of the East China Sea, based on data collected between July 30, 2010 to September 20, 2011 by a surface buoy equipped with an autonomous  $p\text{CO}_2$  system along with hydrological and other chemical sensors. The study site ( $122.37^\circ \text{E}$ ,  $30.55^\circ \text{N}$ ) is influenced by the river plumes of both the Changjiang and Qiantang River and is characterized by strong tidal circulation and highly turbid waters. The amplitude of  $p\text{CO}_2$  changes increased from winter to summer over both diurnal and spring-neap tidal cycle timescales. The average surface water  $p\text{CO}_2$  was slightly undersaturated with respect to the atmosphere in winter ( $382 \pm 18 \mu\text{atm}$ ), but supersaturated in spring ( $500 \pm 56 \mu\text{atm}$ ) and summer ( $687 \pm 110 \mu\text{atm}$ ). Overall the study site was a source of atmospheric  $\text{CO}_2$  with an average sea to air flux of  $14 \pm 9 \text{ mmol C m}^{-2} \text{ d}^{-1}$  from January to October 2011. We revealed factors controlling the  $p\text{CO}_2$  dynamics at different timescales. Over seasonal timescales, temperature and estuarine mixing dominated the seawater  $p\text{CO}_2$  variability. Over spring-neap tidal timescales in winter and spring, the major drivers were similarly water mass mixing and temperature. However, in summer, biological activity and air-sea exchange became the two principal factors controlling the variations in surface seawater  $p\text{CO}_2$ . Our mass balance models further suggested that biological processes impacted surface  $p\text{CO}_2$  differently during different tidal phases. Respiration was revealed to promote the increase in  $p\text{CO}_2$  during spring tide in August, but in neap tides of the same month biological production was evident and resulted in the drawdown of  $p\text{CO}_2$ . This is because photosynthesis was generally limited by light in summer at the study site due to high turbidity, except during neap tides when turbidity was dramatically drawn down, triggering high biological productivity. At the diurnal timescale, sea surface  $p\text{CO}_2$  was primarily controlled by tidal mixing, except during neap tides in summer when sea surface  $p\text{CO}_2$  was greatly influenced by biological metabolism. This study also revealed significant inter-summer differences between 2010 and 2011, showing lower sea surface  $p\text{CO}_2$  in August 2010 as compared to August 2011, which was likely due to the enhanced biological uptake as a result of the relatively low turbidity caused by weak tidal currents and enhanced river flow in August 2010. Our study highlights a highly dynamic system primarily driven by tidal mixing, which not only modulates water mass mixing but also affects turbidity, which subsequently controls biological production. These processes led to a synergy of  $\text{CO}_2$  dynamics in a tidally driven and highly turbid nearshore system, where high frequency time-series observations are essential to reveal the complex controls of  $\text{CO}_2$  dynamics.

## 1. Introduction

Despite their small surface area, nearshore coastal waters and estuarine systems may exhibit the largest dynamics and heterogeneities in the marine realm in terms of both biogeochemical fluxes and their controls, representing a most difficult component of the coastal carbon cycle (Dai et al., 2009; Bauer et al., 2013; Chen et al., 2013; Bakker et al., 2016; Bourgeois et al., 2016). As a consequence, there are much less publicly available  $p\text{CO}_2$  observations in the coastal and estuarine systems compared to open ocean (Bakker et al., 2016). Coastal and estuarine systems are subject to abundant material inputs from land,

thus showing greater temporal variations in  $p\text{CO}_2$  on both intra-seasonal (short-term) and seasonal scales (Borges and Frankignoulle, 1999; Dai et al., 2009; Bauer et al., 2013; Laruelle et al., 2017). For example, Guo et al. (2009) observed a high seasonal variation of  $p\text{CO}_2$ , up to  $\sim 1000 \mu\text{atm}$ , in the highly urbanized upper estuary of the Pearl River. Although some studies have focused on the temporal variability of  $p\text{CO}_2$  in nearshore waters, these prior studies are mostly based on observations that have a high spatial resolution but occur over short timescales or long-term timescales with a low frequency, which makes it difficult to resolve the full spectrum of  $\text{CO}_2$  dynamics across finer timescales and its fluxes at regional or global scales (Borges and Frankignoulle, 1999;

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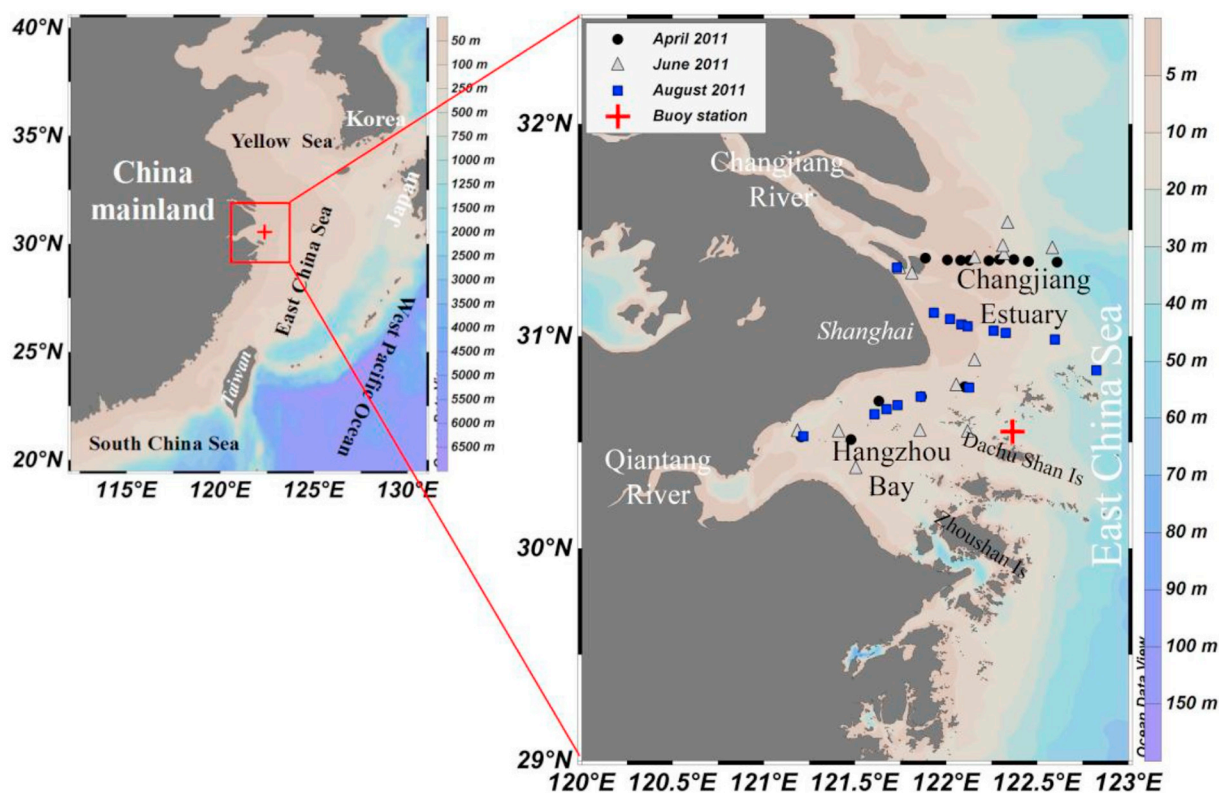


Fig. 1. Map of the East China Sea showing the location of the buoy deployed from August 2010 to September 2011. Also shown are the stations in the Changjiang Estuary, Hangzhou Bay, and inner shelf of the East China Sea sampled during three cruises in April, June, and August 2011. The buoy site was also sampled (the symbols are overlapped). The figure was made with the software of Ocean Data View (Schlitzer, 2013).

Yates et al., 2007; Dai et al., 2009; Schiettecatte et al., 2007).

Mooring platform observations bring the possibility of high frequency  $p\text{CO}_2$  measurements over a range of timescales (Lefèvre et al., 2008; Nemoto et al., 2009; Turk et al., 2010; Xue et al., 2016; Reimer et al., 2017; Li et al., 2018), providing important insights about  $\text{CO}_2$  dynamics that are otherwise difficult to capture. For example, over a one-year period, surface seawater  $p\text{CO}_2$  observations via a moored autonomous  $p\text{CO}_2$  system off the coast of Georgia revealed an abrupt drawdown of  $p\text{CO}_2$  that was not captured by prior ship-based studies (Xue et al., 2016). Along the Georgia coast, temperature is the main driver for the annual cycle of  $p\text{CO}_2$ , although riverine inputs through biological respiration and production also significantly contribute to the variation of seawater  $p\text{CO}_2$  (Xue et al., 2016). In addition, a seven year time-series study in the Bay of Brest, deployed by a CARIOCA sensor on an automatic moored system, have shown that biological processes play a dominant role in the diurnal, seasonal, and intra-annual variability of  $p\text{CO}_2$  (Bozec et al., 2011). These examples clearly demonstrate that the control mechanisms of  $p\text{CO}_2$  in the nearshore waters of different regions under different environmental settings may be highly variable.

Numerous studies have indicated that tidal forcing plays an important role in the biogeochemistry of carbon and other biogenic elements in estuarine systems. For example, tidal height and  $p\text{CO}_2$  inversely correlated in estuaries along the Taiwan Strait, suggesting tidal mixing was the dominant process controlling variations in surface seawater  $p\text{CO}_2$  (Dai et al., 2009). Tidal forcing was observed to significantly affect organic matter exchange between dissolved and particulate pools in tidal estuaries along the Atlantic coast of Europe during both semidiurnal and spring-neap tidal cycles (Middelburg and Herman, 2007). Consistently, the nutrient concentrations in a small California estuary were also predominantly controlled by tidal mixing (Caffrey et al., 2007).

Hangzhou Bay, a subtropical tidal estuary, is located on the inner

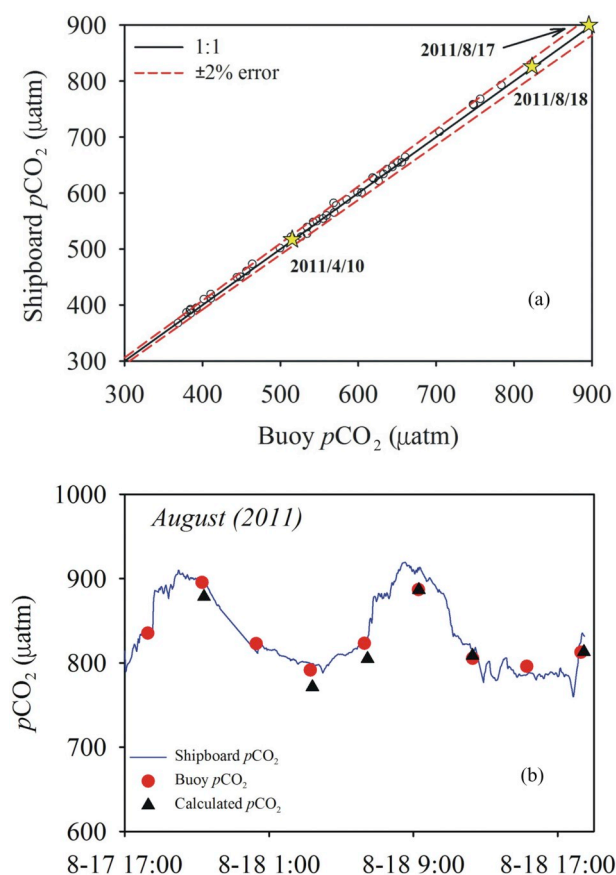
shelf of the East China Sea (ECS), and influenced by the river plumes from both the Qiantang River and Changjiang (or Yangtze River). The Qiantang River and Hangzhou Bay are famous for their macrotidal ranges of up to 8 m during spring tide (Wang, 1990). Due to high turbidity, phytoplankton growth in the system is often limited by light penetration, which can change however based on the extent of seawater intrusion into the estuary (Zhou et al., 2012). Limited studies (Gao et al., 2008; Zhai and Dai, 2009) indicate that this area generally behaves as a source of atmospheric  $\text{CO}_2$ . However, short (diurnal) and long-term (seasonal and inter-annual) monitoring of seawater  $p\text{CO}_2$  in this area are scant (Yu et al., 2013; Tseng et al., 2014; Guo et al., 2015). However, the mechanisms controlling surface seawater  $p\text{CO}_2$  across different timescales have not been thoroughly examined, and an assessment of the response of  $p\text{CO}_2$  to frequently occurring eutrophication events is still lacking.

In this study, we present a large, high frequency data set of  $p\text{CO}_2$  and air-sea  $\text{CO}_2$  fluxes collected during August 2010–September 2011 based on an autonomous buoy system deployed at the mouth of Hangzhou Bay. We examined the main controls of  $p\text{CO}_2$  dynamics over seasonal and neap-spring tidal cycles as well as diurnal timescales. Moreover, we constructed mass balance models to quantify the relative importance of various controlling factors during the spring-tide of summer and in an algal bloom event, which frequently occurs during the summer neap tide. Lastly, we compared monthly  $p\text{CO}_2$  in August 2010 with August 2011 and revealed the possible mechanism modulating the inter-summer changes.

## 2. Material and methods

### 2.1. Study site

Our study site (122.37° E, 30.55° N) is located at the mouth of Hangzhou Bay and has an average water depth of 30 m. It faces the



**Fig. 2.** Underway  $p\text{CO}_2$  measurements versus buoy  $p\text{CO}_2$  observations deployed in western Xiamen Bay from July 7 to July 15, 2010 (black circles, a) and near the buoy site ( $\sim 0.5$  km away) conducted on April 10, August 17 and 18, 2011 (yellow stars, a). Comparison of buoy surface seawater  $p\text{CO}_2$  observations and shipboard time-series measurements of  $p\text{CO}_2$  as well as calculated  $p\text{CO}_2$  from DIC and TA measured near the buoy site ( $\sim 0.5$  km away) from 17:00 August 17 to 17:00 August 18, 2011 (b). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Changjiang Estuary to the north, Hangzhou Bay to the west, and the Zhoushan Islands to the south (Fig. 1). The area is characterized by a strong influence of freshwater inputs from both the Changjiang and Qiantang Rivers, and exhibits estuarine mixing with a salinity range of 20 – 30 (Gao et al., 2011). The Changjiang River is the fourth largest river in the world, with an average freshwater discharge of  $\sim 944 \times 10^9 \text{ m}^3 \text{ yr}^{-1}$  (Dai and Trenberth, 2002). Changjiang River fluxes are greater in summer than in winter, and the structure of the Changjiang plume is seasonally variable. In winter the plume flows southward, confined in a narrow band along the coast. In summer, the main pathway of the plume turns to the northeast, with an extension in a band to the south along the coast (Mao et al., 1963; Beardsley et al., 1985). In addition, because of higher discharge in summer, the Changjiang plume obtains ample kinetic energy to run offshore towards the southeast for a distance and then rotates at  $\sim 122^\circ 30'$  to the northeast (Wu et al., 2011). Consequently, the buoy site is affected by the Changjiang plume throughout the year. Moreover, the Qiantang River could also reach the buoy site after passing through Hangzhou Bay, although the mean Qiantang River discharge is only  $\sim 4\%$  of the Changjiang River (Ni et al., 2003), particularly in summer when most of the Changjiang plume tends to turn northeast after leaving the Changjiang Estuary. Hangzhou Bay and its adjacent area is a highly turbid system; the average suspended particulate matter (TSM) concentrations reach up to  $1.56 \text{ kg m}^{-3}$  (Chen et al., 2006a,b; Xie et al., 2009). TSM concentrations have distinct seasonal and spring-neap tidal

variations, e.g. lower TSM concentrations are generally observed in summer and during the neap tide compared with winter and the spring tide (Chen et al., 2006a,b). This study area featured strong semi-diurnal tidal cycles with a tidal range of 3–4 m (Xie et al., 2009). The velocities of tidal currents during flood and ebb reach  $\sim 80$  and  $\sim 90 \text{ cm s}^{-1}$  respectively and the residual surface current exceeds  $20 \text{ cm s}^{-1}$  (Hu et al., 2009). Therefore, under the impact of strong tides and freshwater from rivers, the biogeochemical processes and carbon cycle at this buoy site might be controlled by tidal mixing. Despite the high turbidity, the mouth of Hangzhou Bay and its adjacent area are frequently reported as sites of red tide/algal blooms in the spring and summer (Zhao et al., 2004; Wang and Wu, 2009).

## 2.2. Time series observations

Data presented in this study were obtained from July 30 to October 20, 2010, and from January 17 to September 20, 2011, by a 10-m-diameter moored buoy. High frequency measurements performed at 3-h intervals of sea surface  $p\text{CO}_2$ , atmospheric  $p\text{CO}_2$  and other related parameters such as sea surface temperature and salinity (SST and SSS), dissolved oxygen (DO), 10-m-wind speed, wind direction, and turbidity were collected during the deployments. An autonomous  $\text{CO}_2$  measurement system, Battelle- $\text{CO}_2$  (© Battelle Memorial Institute), was mounted on the buoy. This Moored Autonomous  $p\text{CO}_2$  (MAPCO2) system combined an “h” shaped bubble air-sea equilibrator with a non-dispersive infrared analyzer Li-820 (LI-COR® Inc.). The Li-820 detector performs two-point calibrations automatically with a zero- $\text{CO}_2$  reference before each sampling interval and  $\text{CO}_2$  standard gases (501.07 ppm in 2010 and 500.94 ppm in 2011) traceable to World Meteorological Organization (WMO) references (Sutton et al., 2014). Aqueous  $p\text{CO}_2$  was estimated as a function of the mole fraction of  $\text{CO}_2$  ( $x\text{CO}_2$ ) in dry air, barometric pressure, and saturated water vapor pressure from Weiss and Price (1980), as well as temperature in the equilibrator headspace and of surface seawater. Detailed calculations and measurement principles are presented in Sutton et al. (2014).

## 2.3. Data quality assessment by shipboard measurements

To assess the data quality of the MAPCO2 system, we compared the buoy data with our homemade underway  $\text{CO}_2$  system (1% accuracy, Zhai et al., 2005a,b) during a time-series investigation in western Xiamen Bay ( $24.5390^\circ \text{N}$ ,  $118.0632^\circ \text{E}$ , from July 7 to July 15, 2010) before the system was mounted on the buoy. A point to point comparison (circles in Fig. 2a) between the two systems shows almost identical results within  $\pm 2\%$  error. In addition, during the deployment of the buoy, we carried out 3 cruise surveys in April, June, and August 2011 in the vicinity of the buoy site (0.3–0.7 km) to verify the buoy data. For the cruises in April and August, a submersible pump was mounted on the ship at a depth of 0.5–2 m below the sea surface, which was generally consistent with the sampling layer ( $\sim 0.5$  m) by the MAPCO2 system. The underway pumping system continuously measured  $\text{CO}_2$  at one-minute intervals. Meanwhile, sea surface temperature and salinity were measured by a YSI® 6600 (YSI® Inc.) at the same interval. Unfortunately, we did not measure sea surface  $p\text{CO}_2$  during the survey in June. In addition, discrete samples for dissolved inorganic carbon (DIC), total alkalinity (TA), pH (NBS buffers,  $25^\circ \text{C}$ ), and DO were also collected from the side vent of the pumping system in April and August, but from Niskin bottles in June. All samples except TA were free of bubbles. DIC and TA were preserved with saturated  $\text{HgCl}_2$  and analyzed within two weeks after sampling. The analytical methods for DIC and TA have been described in Cai et al. (2004) and each has a precision of  $< \pm 2 \mu\text{mol kg}^{-1}$ . The pH at  $25^\circ \text{C}$  was measured on-site with a Corning 350 pH/ion analyzer and a Ross combination electrode (Orions) calibrated against three NIST pH buffers (NBS buffers) at  $25^\circ \text{C}$ . The precision of pH measurements was  $\pm 0.005$  units. DO was measured onboard by the classic Winkler titration method at a precision

of  $\pm 1.25 \mu\text{mol kg}^{-1}$  (0.3%).  $p\text{CO}_2$  calculated by DIC and TA (partly by TA and pH) with the CO2SYS program (Pierrot et al., 2009) was used to further validate the observed  $p\text{CO}_2$  values and fill up the  $p\text{CO}_2$  data gap in June. The dissociation constants of carbonic acid ( $K_1$  and  $K_2$ ) are from Millero et al. (2006). The value of  $\text{KSO}_4^-$  is from Dickson (1990). The concentrations of phosphate and silicate used in the CO2SYS program are  $1.31 \pm 0.12$  and  $45.0 \pm 5.9 \mu\text{mol kg}^{-1}$  ( $n = 6$ ), respectively. The comparison between buoy and shipboard system is still within  $\pm 2\%$  error (Fig. 2a). In addition, hourly time-series observations of  $p\text{CO}_2$  and other related parameters were carried out onboard from 17:00 Beijing Time, 17 August 2011 to 17:00 Beijing Time, 18 August 2011 in proximity to the buoy ( $\sim 0.5$  km) (Fig. 2b). The relative standard deviations among buoy  $p\text{CO}_2$ , shipboard  $p\text{CO}_2$ , and calculated  $p\text{CO}_2$  ranged from 0.6% to 1.8% (average 1.3%), which further demonstrated the good quality of  $p\text{CO}_2$  data obtained from the MAPCO2 system. The average difference between buoy DO (AADERAA® DO) and Winkler DO (figure not shown) measurements was 4%, which agrees with the accuracy of  $< 5\%$  given by AADERAA®. Therefore, we have built a reliable in situ  $p\text{CO}_2$  observation system on a moored buoy with a Battelle- $\text{CO}_2$  system in an estuary of the ECS. These  $p\text{CO}_2$  data will be submitted to SOCAT database (The Surface Ocean  $\text{CO}_2$  Atlas, <https://www.socat.info/>).

## 2.4. Computational methods

### 2.4.1. Temperature effect on the variation of $p\text{CO}_2$

To remove the temperature effect, aqueous  $p\text{CO}_2$  was normalized to mean sea surface temperature (SST) following Takahashi et al. (1993, 2002):

$$p\text{CO}_{2\_nontherm} = p\text{CO}_2 \times e^{0.0423 \times (\overline{\text{SST}} - \text{SST})} \quad (1)$$

where  $p\text{CO}_{2\_nontherm}$  is the observed  $p\text{CO}_2$  after temperature normalization, representing  $p\text{CO}_2$  without a temperature effect. SST is the in situ sea surface temperature and  $\overline{\text{SST}}$  represents the mean value during the target period.

The thermal forcing on aqueous  $p\text{CO}_2$  can be estimated as in Dai et al. (2009),

$$p\text{CO}_{2\_therm} = p\text{CO}_2 - p\text{CO}_{2\_nontherm} = p\text{CO}_2 - p\text{CO}_2 \times e^{0.0423 \times (\overline{\text{SST}} - \text{SST})} \quad (2)$$

where  $p\text{CO}_{2\_therm}$  represents aqueous  $p\text{CO}_2$  values under the influence of temperature.

### 2.4.2. Air-sea $\text{CO}_2$ exchange flux

The air-sea  $\text{CO}_2$  flux (F) was calculated as

$$F = k \times \alpha \times \Delta p\text{CO}_2 = k \times \alpha \times (p\text{CO}_{2\_seawater} - p\text{CO}_{2\_air}) \quad (3)$$

where  $k$  is the  $\text{CO}_2$  transfer velocity,  $\alpha$  is  $\text{CO}_2$  solubility (Weiss, 1974), and  $\Delta p\text{CO}_2$  is the partial pressure difference of  $\text{CO}_2$  between surface seawater and the atmosphere. A positive flux value represents a net  $\text{CO}_2$  exchange from the water body to the air.  $\text{CO}_2$  transfer velocity as a function of wind speed is determined from Sweeney et al. (2007) as follows,

$$k = f \times u_{10}^2 \times (660/Sc)^{0.5} \quad (4)$$

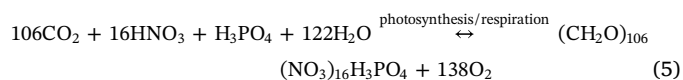
where,  $f = 0.27$ , and  $u_{10}$  is the wind speed at 10 m above the sea surface.  $Sc$  is the Schmidt number as a function of temperature (Wanninkhof, 1992).

### 2.4.3. Quantification of factors influencing the surface water $p\text{CO}_2$

We constructed a one-dimensional budget model to evaluate various  $p\text{CO}_2$  controlling factors and their uncertainties according to Xue et al. (2016), with the details presented in Appendix S1 and S2.

### 2.4.4. Biology effect on the variation of $p\text{CO}_2$

The main biological processes consuming and generating  $\text{CO}_2$  were illustrated as follows,



According to Eq. (5), we calculated the change of DIC ( $\delta\text{DIC}$ ) due to biological effects and the change of  $p\text{CO}_2$  ( $\delta p\text{CO}_2$ ) was estimated by the Revelle Factor ( $\beta$ ) (Revelle and Suess, 1957) as,

$$\beta = (\delta p\text{CO}_2 / p\text{CO}_2) / (\delta\text{DIC} / \text{DIC}) \quad (6)$$

$$\delta p\text{CO}_2 = \beta \times \delta\text{DIC} / \overline{\text{DIC}} \times \overline{p\text{CO}_2} \quad (7)$$

where  $\overline{\text{DIC}}$  and  $\overline{p\text{CO}_2}$  represent the average values of each parameter. In this study,  $\beta$ , which is the buffer (Revelle) factor, is  $\sim 15.3$ , calculated using the CO2SYS program from TA and DIC sampled at the buoy site in August (Pierrot et al., 2009). DIC was set to  $2000 \mu\text{mol kg}^{-1}$ ; a reasonable range in its values would have a minor influence on  $p\text{CO}_2$  fluctuations.

## 3. Results

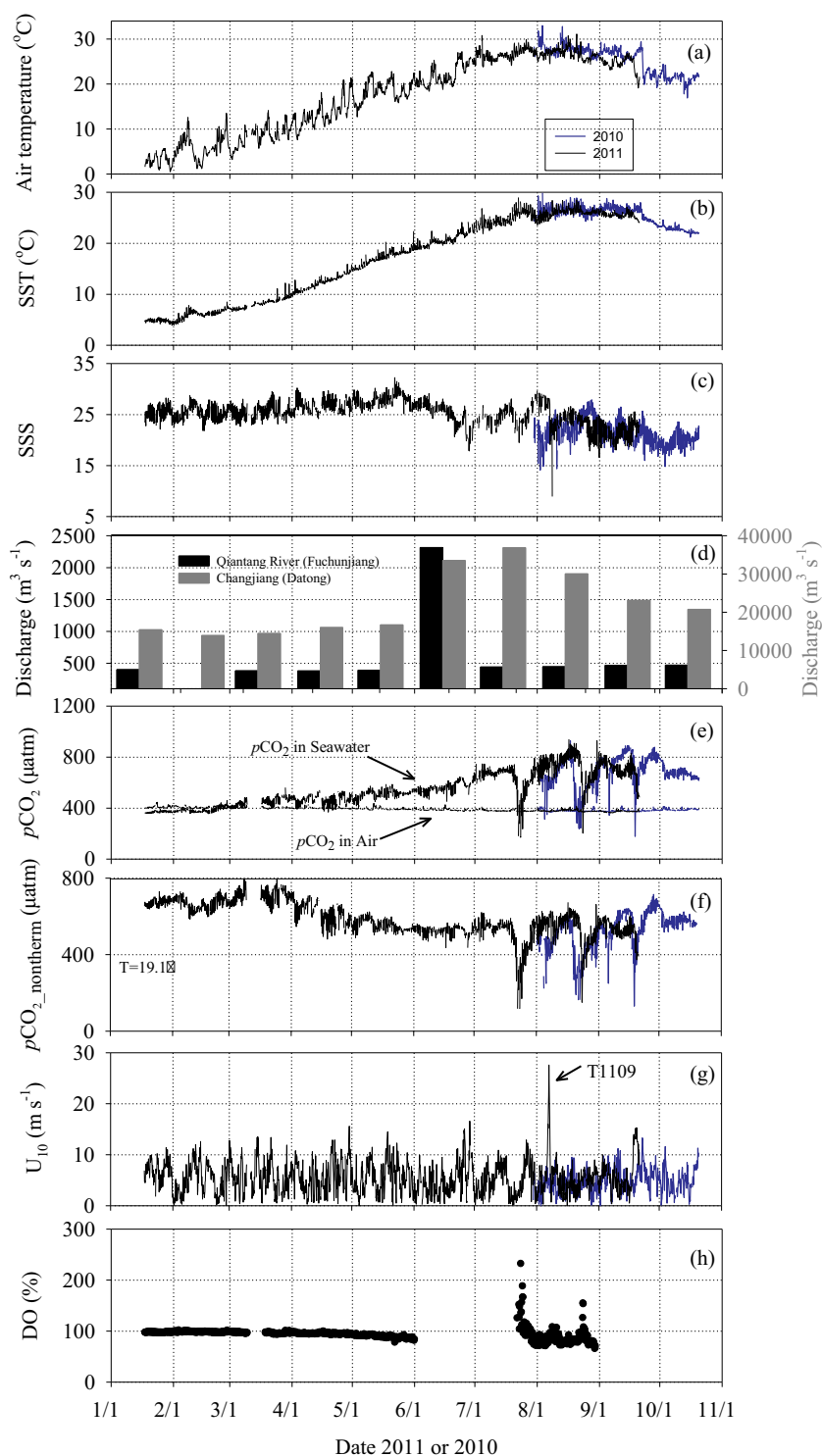
### 3.1. Data overview

Shown in Fig. 3 is the time-series data of air temperature, SST, SSS, atmospheric  $p\text{CO}_2$ , sea surface  $p\text{CO}_2$ ,  $p\text{CO}_{2\_nontherm}$  (normalized to  $19.1^\circ\text{C}$ ), wind speed at height of 10 m above sea surface, and DO saturation via the moored buoy from July 30–October 20, 2010, and January 17–September 20, 2011. Data are not available from October 21, 2010 to January 16, 2011 due to maintenance of the buoy. This unique dataset thus covers two summers, part of fall, one winter and one spring season and allows us to examine timescale variability and its controls. It should be noted that the turbidity data are available and reliable solely in summer (from July to September 2011), and the details are presented in Section 4.3.

SST gradually increased from January ( $4\text{--}5^\circ\text{C}$ ) to August ( $25\text{--}30^\circ\text{C}$ , Fig. 3b), consistent with the trend in air temperature (Fig. 3a), suggesting that SST was mainly controlled by seasonal warming. Based on these temperature variations, time-series data in this study were divided into three stages/seasons: a low temperature period/winter (January and February), a period of rising temperatures in spring (March to June), and a high temperature period in summer (July to September).

SSS generally had a negative relationship with river discharge. Low SSS corresponded to increased river discharge in summer, suggesting a strong terrestrial influence during this season. However, the variation of mean SSS was not perfectly consistent with river discharge: e.g. from January to May SSS increased gradually from a monthly average of 25.4 to 27.9 (Fig. 3c), despite the Changjiang discharge being constant ( $\sim 15,000 \text{ m}^3 \text{ s}^{-1}$ ) during these periods (Fig. 3d). Additionally, the lowest SSS existed in September, but the highest river flows occurred in June and July (Fig. 3c and d). This discrepancy indicated that the river plume did not fully reach the location of the study site during all seasons. In other words, SSS at the buoy was not solely related to riverine discharge, but also likely affected by large seasonal variations in river plume paths as described in Section 2.1. The monthly discharge of the Changjiang was 14–83 fold greater than the Qiantang River in 2011 (Fig. 3d).

Over the entire study period, sea surface  $p\text{CO}_2$  ranged from 170 to  $940 \mu\text{atm}$ , generally arose from winter, spring, to summer (Fig. 3e). The surface water  $p\text{CO}_2$  exhibited remarkably large variability with largest amplitudes in summer when it also occurred the maximum and minimum values (Fig. 3e). Atmospheric  $p\text{CO}_2$  showed relatively stable

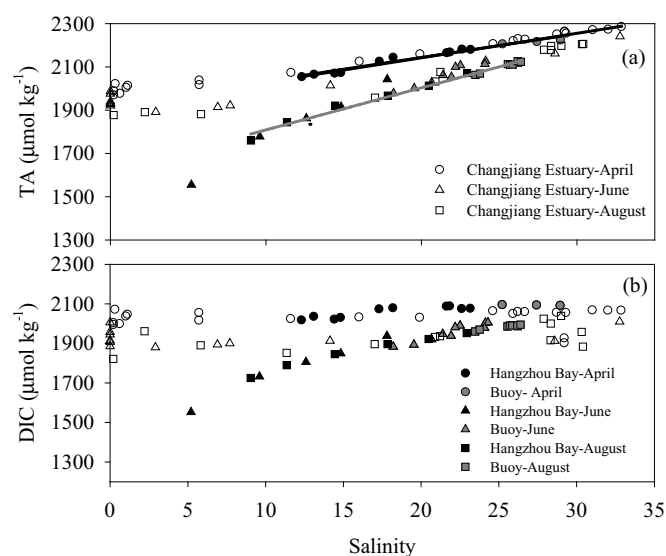


**Fig. 3.** Time-series observations of atmospheric temperature (a), sea surface temperature (b), sea surface salinity (c), monthly discharges of the Changjiang (Datong Gauge station) and Qiantang River (Fuchunjiang Gauge station) (d), sea surface  $p\text{CO}_2$  (e), temperature-normalized sea surface  $p\text{CO}_2$  ( $p\text{CO}_{2\_nontherm}$ , normalized to 19.1 °C) (f), wind speed at height of 10 m above sea surface (g), and dissolved oxygen (DO) saturation (h) at the buoy from August 2010 to September 2011. Blue and black lines represent data from 2010 and 2011, respectively. The high wind speeds were observed on August 6 and 7, 2011 with the influence of a typhoon, MUIFA, T1109 (g). Note that data are lacking from October 21, 2010 to January 16, 2011 due to the buoy's maintenance work. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

seasonal variations, with a maximum value of  $407 \pm 8 \mu\text{atm}$  in winter and a minimum value of  $379 \pm 7 \mu\text{atm}$  in summer (Fig. 3e), which agrees well with seasonal fluctuations of atmospheric  $\text{CO}_2$  background values measured on land ( $30^\circ 18' \text{ N}$ ,  $119^\circ 44' \text{ E}$ , 250 km away from our study site) (Pu et al., 2011). The average atmospheric  $p\text{CO}_2$  value during the study period in 2011 was  $392 \pm 14 \mu\text{atm}$ , which was in consonance with the global level that year (Le Quéré et al., 2018). It should be noted that sea surface  $p\text{CO}_2$  started to decrease in October

2010 compared to summer, but due to lack of sufficient data this study did not intend to assess the variability of  $p\text{CO}_2$  and its controlling factors in fall. When the temperature effect was removed, the seasonal increase trend disappeared. The highest  $p\text{CO}_{2\_nontherm}$  existed around winter from January to the early of April, but the variability pattern of  $p\text{CO}_{2\_nontherm}$  kept consistent with  $p\text{CO}_2$  in summer, and the minimum  $p\text{CO}_{2\_nontherm}$  still occurred in summer.

There are no significant monthly variations in wind speed except on



**Fig. 4.** Total alkalinity (TA) versus salinity (a) and dissolved inorganic carbon (DIC) versus salinity (b) in the Changjiang Estuary, Hangzhou Bay, and at the buoy site sampled during April, June, and August 2011. The straight lines in Panel (a) indicate conservative mixing in spring (April) and summer (June and August), respectively.

August 6 and August 7, 2011, when the wind speed was very high due to the influence of a typhoon (MUIFA, T1109, Fig. 3g).

DO saturation state was constant at around 100% from January 17 to May 31, 2011, but ranged from 66% to 232% during July 22 to August 22, 2011.

TA generally behaves conservatively, which makes it a plausible tracer of water mass mixing (Zhai et al., 2007). TA correlated well with salinity in Hangzhou Bay, including at the buoy site, during the cruises conducted in April, June, and August of 2011 (Fig. 4a). In April, TA in Hangzhou Bay and the Changjiang Estuary generally followed the same mixing line. Considering the much greater discharge of the Changjiang River compared to the Qiantang River (~30 fold higher) in April (Fig. 3d), during this time the buoy site was dominated by mixing of two water masses from the Changjiang plume and ECS. However, in June and August the TA mixing line in Hangzhou Bay largely deviated from the Changjiang Estuary, suggesting a significant contribution from the Qiantang River (discharge ~2316 m s<sup>-1</sup> in June) to the buoy site in summer. This may be due to the fact that the core of the Changjiang plume turns northeast after leaving the Changjiang Estuary in summer when it can hardly influence the buoy site.

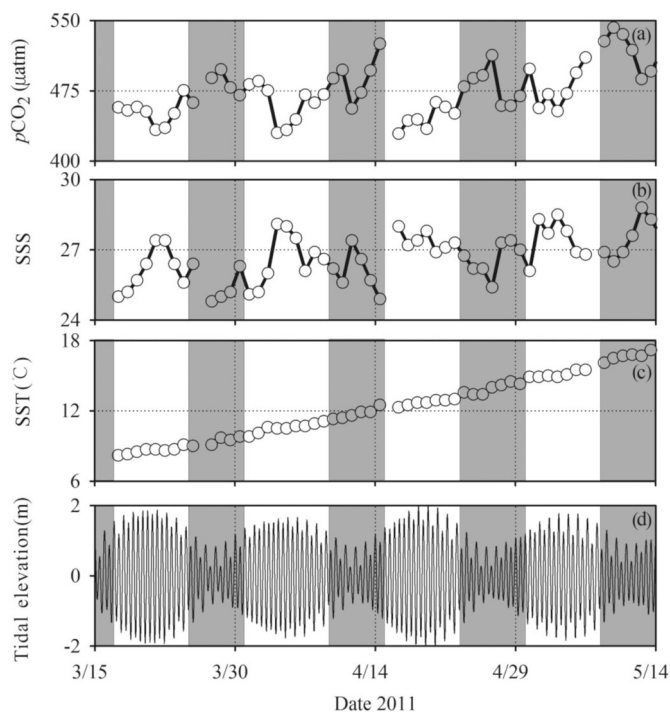
### 3.2. Seasonal variations of surface seawater pCO<sub>2</sub>

The average pCO<sub>2</sub> and related parameters in different seasons are shown in Table 1. It should be noted that we only used the 2011 data to calculate the average seasonal values. The average SST increased from winter to spring to summer, with values of 5.7 ± 0.9 °C, 15.0 ± 4.8 °C, and 25.5 ± 1.1 °C, respectively. SSS during these three seasons was 25.4 ± 1.4, 26.5 ± 1.9, and 23.5 ± 2.5, respectively. Tidal range was constant during these seasons with values of 2.90 ± 0.8 m in winter, 2.78 ± 0.8 m in spring, and 2.80 ± 0.8 m in summer. Sea surface pCO<sub>2</sub> showed large seasonal changes that followed temperature variations, with the highest pCO<sub>2</sub> values observed in summer (687 ± 110 μatm), the lowest in winter (382 ± 18 μatm), and medium values in the spring (500 ± 56 μatm). The overall range in average pCO<sub>2</sub> values among seasons was 305 ± 111 μatm (Fig. 7). Average DO saturation levels were nearly saturated in winter (98% ± 1%) and spring (94% ± 4%), but undersaturated and variable in summer (89% ± 20%).

**Table 1** Seasonal variations in sea surface salinity (SSS), sea surface temperature (SST), tidal range, dissolved oxygen (DO) saturation, pCO<sub>2</sub> in surface seawater, atmospheric pCO<sub>2</sub>, wind velocity, and air-sea CO<sub>2</sub> exchange fluxes at the study site.

Period	Season	SSS	SST (°C)	Tidal range (m)	DO (%)	Sea surface pCO <sub>2</sub> (μatm)	Atmospheric pCO <sub>2</sub> (μatm)	Wind velocity (m s <sup>-1</sup> )	CO <sub>2</sub> flux <sup>a</sup> (mmol m <sup>-2</sup> d <sup>-1</sup> )
2011/1/18-2/28 (41 days)	Winter	25.4 ± 1.4	5.7 ± 0.9	2.90 ± 0.8	98 ± 1	382 ± 18	407 ± 8	5.7 ± 2.9	-3 ± 2
2011/3/1-6/30 (108 days)	Spring	26.5 ± 1.9	15.0 ± 4.8	2.78 ± 0.8	94 ± 4	500 ± 56	396 ± 10	5.5 ± 3.1	10 ± 13
2011/7/1-9/20 (82 days)	Summer	23.5 ± 2.5	25.5 ± 1.1	2.80 ± 0.8	89 ± 20	687 ± 110	379 ± 7	5.4 ± 3.3	30 ± 27

Error bars represent standard deviations of temporal variability.  
<sup>a</sup> gas transfer velocity of CO<sub>2</sub> was calculated from Sweeney et al. (2007).



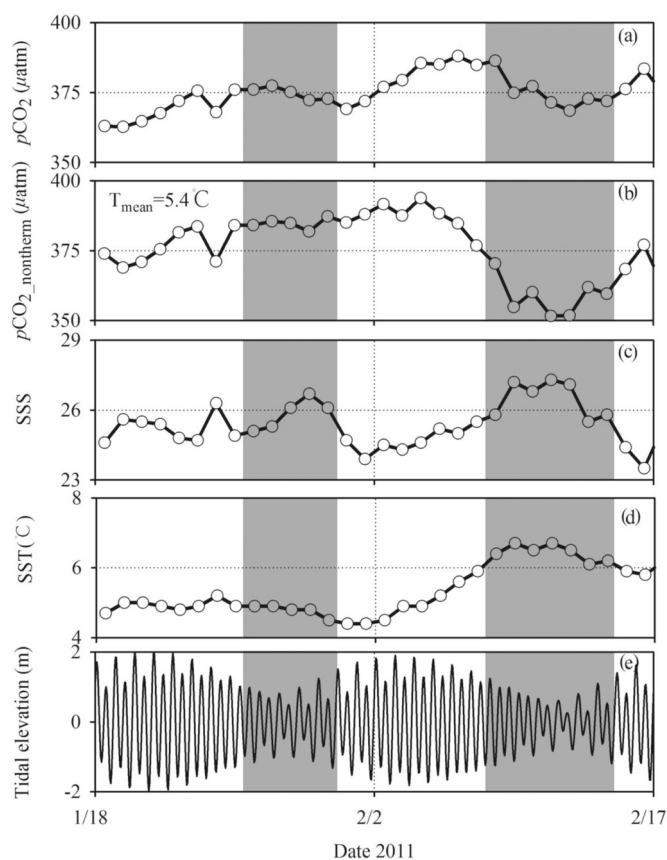
**Fig. 5.** Variations in daily-averaged sea surface  $p\text{CO}_2$  (a), sea surface salinity (SSS) (b), sea surface temperature (SST) (c), and tidal elevation (d) in spring (March 15 to May 14, 2011) at the mooring site. Shaded areas denote neap tide periods.

### 3.3. Surface seawater $p\text{CO}_2$ variability of intra-seasonal scales (spring-neap/neap-spring tide)

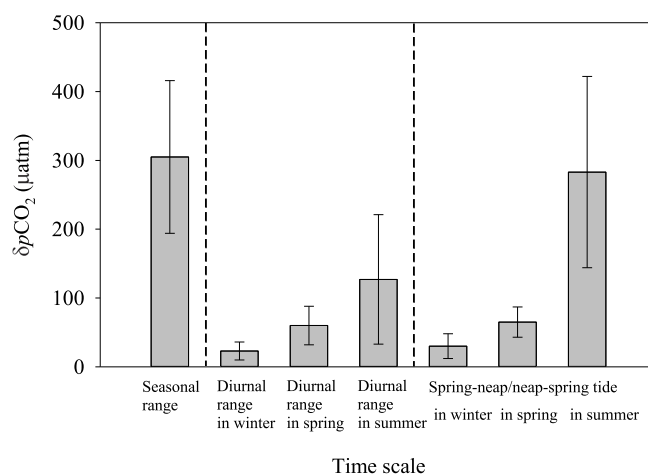
Variations in daily-averaged SSS in spring and summer were roughly larger during spring tides and smaller during neap tides (Fig. 5b; Appendix Figs. S1c, S2b), although this was not the case in winter (Fig. 6c). Daily-averaged sea surface  $p\text{CO}_2$  in spring together with temperature-normalized  $p\text{CO}_2$  ( $p\text{CO}_{2,\text{nontherm}}$ ) in winter generally had an inverse relationship with SSS (Fig. 5a, b; Fig. 6b, c). Large variability of daily-averaged SST existed in spring ( $\sim 9^\circ\text{C}$ ) as compared to that in winter ( $\sim 2^\circ\text{C}$ ) and summer ( $3\text{--}4^\circ\text{C}$ ) (Figs. 5c, 6d; Appendix Figs. S1d, S2c). Small variations in daily-averaged sea surface  $p\text{CO}_2$  were observed in winter, with amplitudes in spring-neap/neap-spring cycles ranging from 15 to 57  $\mu\text{atm}$  (average  $30 \pm 18 \mu\text{atm}$ ) (Fig. 7). In spring, they ranged from 35 to 96  $\mu\text{atm}$  (average  $65 \pm 22 \mu\text{atm}$ ) (Fig. 7). In summer, daily-averaged sea surface  $p\text{CO}_2$  showed distinct declines during neap tides, with the minima of  $p\text{CO}_2$  reaching below the level in the air (Appendix Figs. S1a, S2a). The largest  $p\text{CO}_2$  amplitudes were observed in summer, ranging from 52 to 480  $\mu\text{atm}$  with an average of  $283 \pm 139 \mu\text{atm}$  (Fig. 7), similar to the average seasonal variation range of  $305 \pm 111 \mu\text{atm}$  (Fig. 7).

### 3.4. Diurnal variability of surface seawater $p\text{CO}_2$

The study site showed a regular semidiurnal tide, and the daily tidal range decreased from  $\sim 4$  m at spring tides to  $\sim 2$  m at neap tides. The diurnal amplitude of SST in winter, spring, and summer was from 0.5 to  $1.2^\circ\text{C}$ , 0.2 to  $1.6^\circ\text{C}$ , and 0.2 to  $2.6^\circ\text{C}$ , respectively. Variations in SSS generally correlated with tidal cycles, with lower values observed during low tides and higher values coinciding with high tides (Appendix Fig. S3b, c, Fig. S4b, c, and Fig. S5b, d). Sea surface  $p\text{CO}_2$  was significantly inversely correlated with SSS in both winter and spring (Appendix Fig. S3a, b, and Fig. S4a, b). Diurnal variability of  $p\text{CO}_2$  during the spring tide in summer followed the same trend as in winter and spring, while it tended to be consistent with day-night cycles

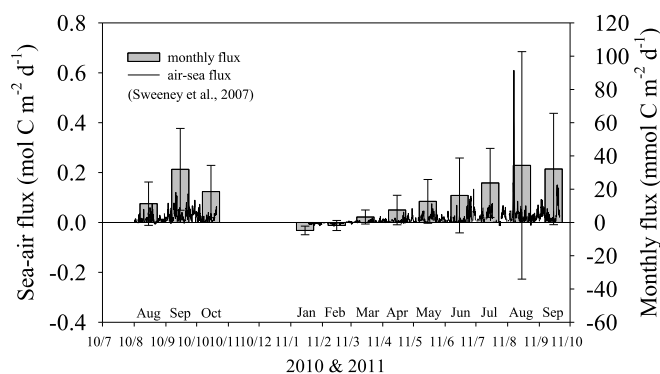


**Fig. 6.** Variations in daily-averaged sea surface  $p\text{CO}_2$  (a), temperature-normalized sea surface  $p\text{CO}_2$  ( $p\text{CO}_{2,\text{nontherm}}$ ) (b), sea surface salinity (SSS) (c), sea surface temperature (SST) (d), and tidal elevation (e) in winter (January 18 to February 17, 2011) at the mooring site. Gray shaded areas represent neap tide periods.



**Fig. 7.** Variation in ranges of sea surface  $p\text{CO}_2$  ( $\delta p\text{CO}_2$ ) at the seasonal, diurnal, and spring-neap/neap-spring scale in winter, spring, and summer. Error bars are standard deviations, which characterize the temporal variability rather than analytical errors in the measurement of sea surface  $p\text{CO}_2$ .

during neap tides, with the minima occurring at  $\sim 18:00$  h (Beijing Time) and generally rising at night (Appendix Fig. S5a, b). The diurnal amplitudes of sea surface  $p\text{CO}_2$  in winter, spring, and summer ranged from 6 to 56  $\mu\text{atm}$  (average  $23 \pm 13 \mu\text{atm}$ ) (Fig. 7), 15 to 162  $\mu\text{atm}$  ( $60 \pm 28 \mu\text{atm}$ ) (Fig. 7), and 16 to 636  $\mu\text{atm}$  (average  $127 \pm 94 \mu\text{atm}$ )



**Fig. 8.** Daily and monthly sea-air  $\text{CO}_2$  fluxes at the mooring site from August to October 2010, and January to September 2011. Positive values represent  $\text{CO}_2$  degassing. The monthly flux equaled the average value from the first day to last day of the month with the exception of October (from 1<sup>st</sup>-20<sup>th</sup>, 2010), January (from 17<sup>th</sup>-31<sup>st</sup>, 2010), and September (from 1<sup>st</sup>-20<sup>th</sup>, 2011). Error bars are standard deviations, which characterize the temporal variations in fluxes.

(Fig. 7), respectively. Summer had much larger diurnal variations in  $p\text{CO}_2$  relative to winter and spring (Fig. 7).

## 4. Discussion

### 4.1. Air-sea $\text{CO}_2$ exchange flux

Sea surface  $p\text{CO}_2$  was above atmospheric  $p\text{CO}_2$  levels most of the time (from March to September 2011 and August to October 2010), with the exception of winter and part of the summer (Fig. 3e). It is thus clear that the study site was overall a source of  $\text{CO}_2$  for the atmosphere, with an average air-sea flux of  $14 \pm 9 \text{ mmol C m}^{-2} \text{ d}^{-1}$  from January to October 2011. In winter, the study area was a sink of atmospheric  $\text{CO}_2$  with an average air-sea flux of  $-4.8 \pm 3 \text{ mmol C m}^{-2} \text{ d}^{-1}$  in January and  $-1.8 \pm 3 \text{ mmol C m}^{-2} \text{ d}^{-1}$  in February (Fig. 8). Although the sharp drop of surface water  $p\text{CO}_2$  appeared in summer (July and August 2011, August and September 2010), the average air-sea flux was  $\sim 3$ -fold higher in summer vs. in spring of 2011 ( $30 \pm 26$  vs.  $9.9 \pm 7 \text{ mmol C m}^{-2} \text{ d}^{-1}$ ). Since there are no significant monthly variations in wind speed, the monthly variability of air-sea flux is dominated by the  $\Delta p\text{CO}_2$  (difference between surface seawater  $p\text{CO}_2$  and atmospheric  $p\text{CO}_2$ ).

### 4.2. Controls on seasonal variations in surface water $p\text{CO}_2$

Sea surface  $p\text{CO}_2$  in nearshore systems is modulated by a variety of factors including temperature, tidal mixing and biological activity, etc. Temperature affects sea surface  $p\text{CO}_2$  through changing the thermodynamic conditions of the carbonate system, and tides determine sea surface  $p\text{CO}_2$  through mixing of different water masses. Biological processes may include photosynthesis, respiration and calcium carbonate precipitation or dissolution, which alter the mass balance of the carbonate system and thus affect  $p\text{CO}_2$  in seawater (Dai et al., 2009).

Temperature typically dominates the seasonal variations in sea surface  $p\text{CO}_2$  in many previous studies (Bates et al., 1998; Nemoto et al., 2009; Turk et al., 2010; Lu et al., 2011). In this study, thermally affected  $p\text{CO}_2$  ( $p\text{CO}_{2\text{therm}}$ ) was generally consistent with the distribution of  $p\text{CO}_2$  (Fig. 9a) except in summer, clearly suggesting that temperature was a principal factor in controlling the seasonal variations in  $p\text{CO}_2$ . A large deviation between measured  $p\text{CO}_2$  and thermally affected  $p\text{CO}_2$  was observed in some days in summer (July, August, and September), and this dramatic drawdown might be influenced by strong biological uptake. Additionally, horizontal mixing could be another important driving factor leading to seasonal variations in  $p\text{CO}_2$ , which could be evidenced by a strong correlation between temperature-normalized

$p\text{CO}_2$  ( $p\text{CO}_{2\text{nontherm}}$ ) and sea surface salinity ( $p < .0001$ , Fig. 9b) in winter (January, February) and spring (March, April, and May). Exceptions existed in summer, as insignificant relationships were observed in June ( $R^2 = 0.003$ ,  $p = .48$ ) and July ( $R^2 = 0.09$ ,  $p = .47$ ) of 2011 (Fig. 9b). In addition, winter had higher  $p\text{CO}_{2\text{nontherm}}$  values compared to summer (Figs. 3f and 9b). The estimated  $p\text{CO}_2$  in bottom water was higher than that at the surface water in August (Appendix Fig. S6). Therefore, the greater  $p\text{CO}_{2\text{nontherm}}$  in winter might be explained by  $\text{CO}_2$ -enriched bottom water mixing with surface water due to winter cooling and monsoonal winds, together with strong biological activity in summer. The measured DIC concentration in April was higher than that in June and August at the mooring site (Fig. 4b), which also support the notion.

### 4.3. Controls on intra-seasonal variations in surface water $p\text{CO}_2$

The amplitude of  $p\text{CO}_2$  induced by temperature ( $\delta p\text{CO}_{2\text{therm}}$ ) was minor compared to non-temperature affected  $p\text{CO}_2$  amplitudes ( $\delta p\text{CO}_{2\text{nontherm}}$ ) in the summers of 2010 and 2011 (Fig. 10c, d). However, in winter and spring, temperature could be one important factor that changes surface water  $p\text{CO}_2$  (Fig. 10a, b).

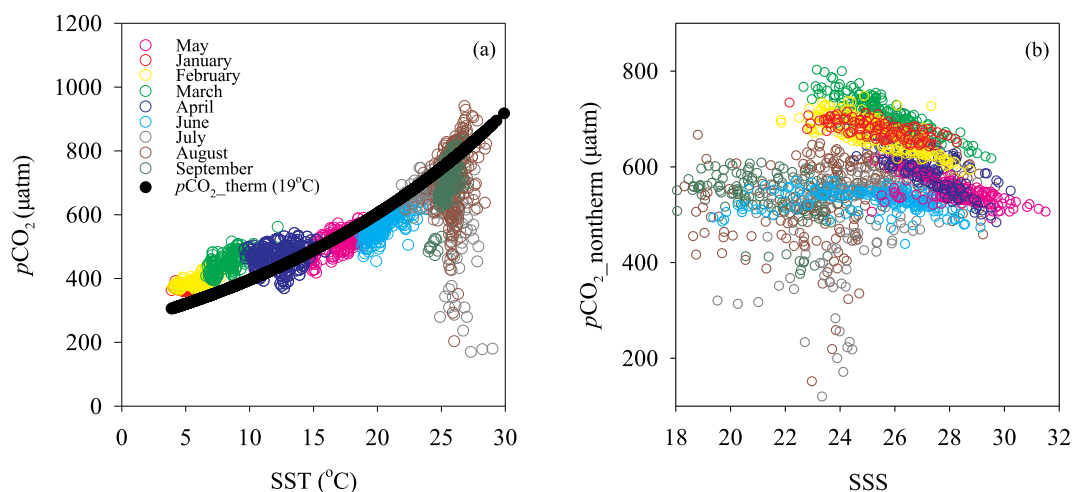
The generally mirrored relationship found between  $p\text{CO}_2$  and SSS in winter and spring demonstrated that mixing of water masses may have contributed greatly to the variations in  $p\text{CO}_2$  of these two seasons (Figs. 6b, c and 5a, b). The positive correlations between  $p\text{CO}_{2\text{nontherm}}$  and DO in winter and spring (Fig. 11) suggested that biological metabolism played a minor role in changing  $p\text{CO}_2$ . In winter, both  $p\text{CO}_{2\text{nontherm}}$  and DO have good correlations with SSS (Fig. 9b, Appendix Fig. S7), and the ratio between the slope of  $p\text{CO}_{2\text{nontherm}}$  vs. SSS ( $-8.0$ ) and slope of DO vs. SSS ( $-5.7$ ) was around 1.4, which was comparable with the slope between  $p\text{CO}_{2\text{nontherm}}$  and DO (1.44), indicating that water masses mixing dominated the variations of  $p\text{CO}_{2\text{nontherm}}$  and DO in winter.

To quantify the various  $p\text{CO}_2$  controlling factors, we constructed a mass balance model based on Xue et al. (2016), with the details presented in Appendix S1. We applied the budget model to April 3–30 which represents spring. Fig. 12 shows the overall  $p\text{CO}_2$  change and contributions from temperature effects, air-sea exchange, water mass mixing, biological metabolism, and residual term (mainly reflecting the sum of contributions of any factors not listed, such as diffusion, carbonate mineral dissolution/formation) during the target periods. In April, temperature was the predominant controlling factor affecting  $p\text{CO}_2$ , leading to a  $85 \pm 9 \mu\text{atm}$  rise in  $p\text{CO}_2$  as the water temperature increased by  $4.2^\circ\text{C}$ .  $p\text{CO}_2$  variations as a function of temperature in April matched well with the  $p\text{CO}_{2\text{therm}}$  curve (Fig. 9a). Moreover, horizontal mixing between nearshore and offshore water also played a significant role in contributing to  $p\text{CO}_2$  variability ( $41 \pm 14 \mu\text{atm}$ ). We assumed the salinity differences were solely attributed to mixing processes, as a strong correlation ( $R^2 = 0.66$ ,  $P < .0001$ ) observed between  $p\text{CO}_{2\text{nontherm}}$  and salinity in April (Fig. 9b) validates the importance of water mass mixing. In summary, temperature, mixing, and air-sea exchange were the major controlling factors, contributing respectively 51%, 20%, and  $-20\%$  to the absolute  $p\text{CO}_2$  change in April.

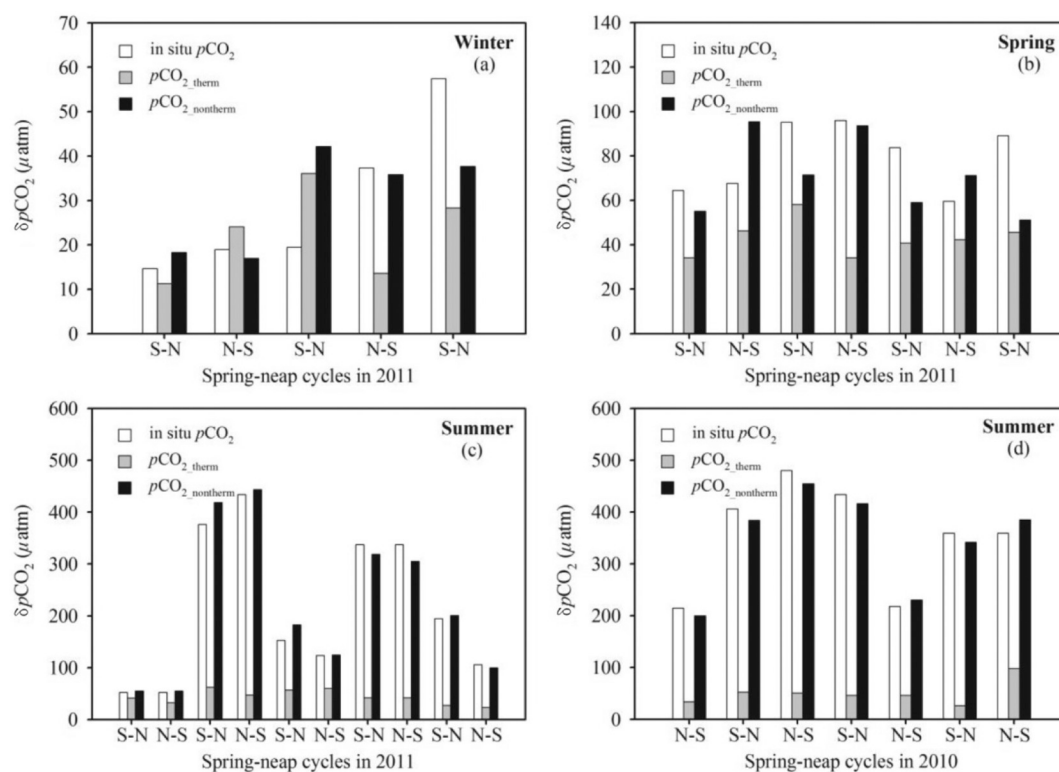
In spring, significant positive correlations occurred between daily-averaged  $p\text{CO}_{2\text{nontherm}}$  and DO (Fig. 11) with a slope 1.8. Like in winter, strong correlations also existed between  $p\text{CO}_{2\text{nontherm}}$  vs. SSS, and DO vs. SSS in spring (Appendix Fig. S7). However, the ratio between the slope of  $p\text{CO}_{2\text{nontherm}}$  vs. SSS ( $-37.0$ ) and slope of DO vs. SSS ( $-12.1$ ) was 3.0, higher than the slope of 1.8. This discrepancy might be explained by the different effect of air-sea exchange on the variability of  $p\text{CO}_{2\text{nontherm}}$  and DO in spring. Therefore, although we observed positive correlations between  $p\text{CO}_{2\text{nontherm}}$  and DO in both spring and winter (Fig. 11), water masses mixing and air-sea exchange drive the variation of  $p\text{CO}_{2\text{nontherm}}$  in spring; in winter the  $p\text{CO}_{2\text{nontherm}}$  is mainly controlled by water masses mixing.

In summer, a negative correlation was found between  $p\text{CO}_{2\text{nontherm}}$





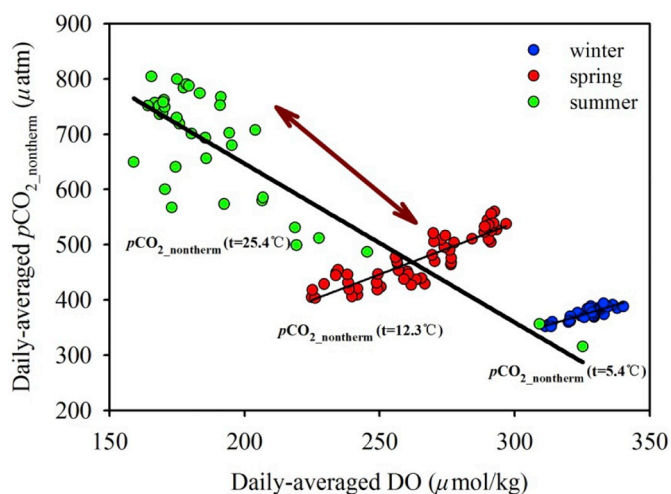
**Fig. 9.** Sea surface  $p\text{CO}_2$  versus sea surface temperature (SST) (a), temperature-normalized sea surface  $p\text{CO}_2$  ( $p\text{CO}_{2\_nontherm}$ , normalized to 19.1 °C) versus sea surface salinity (SSS) (b) in 2011. A strong correlation between  $p\text{CO}_{2\_nontherm}$  and SSS in winter ( $p < .0001$ , January and February) and spring ( $p < .0001$ , March, April, and May). However, insignificant relationships were observed in June ( $p = .48$ ) and July ( $p = .47$ ) of 2011.



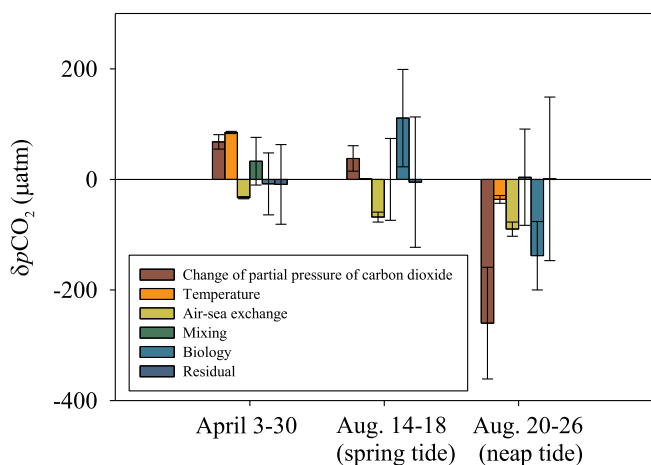
**Fig. 10.** The amplitude of sea surface  $p\text{CO}_2$ ,  $p\text{CO}_{2\_therm}$ , and  $p\text{CO}_{2\_nontherm}$  during spring-neap and neap-spring tidal cycles in winter (January 18 to February 28) (a), spring (March 15 to May 14) (b), summer 2011 (July 1 to September 21) (c), and summer 2010 (August 1 to September 30) (d) at the mooring site.

and DO concentration, with a Pearson coefficient of 0.84 ( $P < .1$ , Fig. 11) and a slope of  $-2.9$ , which means a  $1 \mu\text{mol kg}^{-1}$  increase in DO can reduce  $p\text{CO}_2$  by  $\sim 3 \mu\text{atm}$ . This is comparable with the organic metabolism effect, of which a  $1 \mu\text{mol kg}^{-1}$  variation of DO can cause an  $\sim 4 \mu\text{atm}$  inverse variation in  $p\text{CO}_2$ . Consequently, photosynthesis and respiration might be the predominant mechanism controlling the spring-neap variations in  $p\text{CO}_2$  in summer. Daily-averaged sea surface turbidity observed at the study site in summer (2011) ranged from 5 to 220 NTU, which corresponded to 6–320  $\text{mg L}^{-1}$  of TSM in terms of the relationship reported by He et al. [He et al., 2013, TSM ( $\text{mg L}^{-1}$ ) =  $-0.72 + 1.45 \times \text{turbidity (NTU)}$ ]. High concentrations of suspended particles lead to high turbidity and low light transmittance,

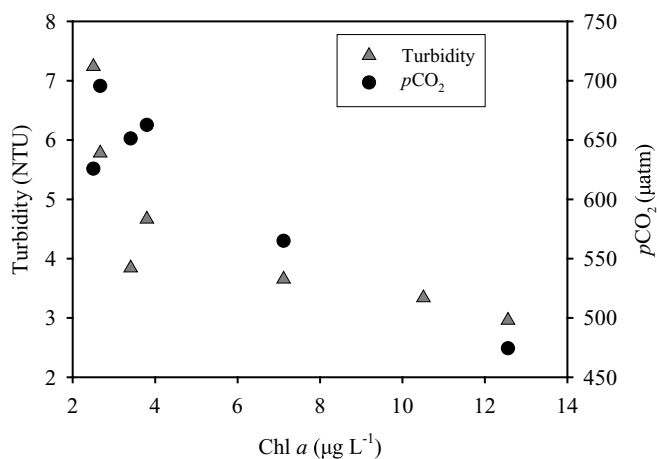
and thus photosynthetic production was likely to be limited by light (Ning et al., 1988; Cai et al., 2002). In general, TSM concentrations were higher in winter than in summer (Chen et al., 2006a,b) because in summer resuspension is weakened by increasing riverine discharge. Chen et al. (2006a,b) conclude that tide-induced sediment resuspension ultimately controls the concentration of TSM. It has also been found that photosynthesis is enhanced when the concentration of TSM is  $< 10 \text{ mg L}^{-1}$  in the plumes of the Changjiang and Amazon River (DeMaster et al., 1986; Ning et al., 1988; DeMaster and Pope, 1996). At our study site, we did not measure Chl *a* during 2010–2011. Instead, we used the sensor data of Chl *a*, turbidity (a FLNTU Combination Fluorometer and Turbidity Sensor) and  $p\text{CO}_2$  deployed in another buoy



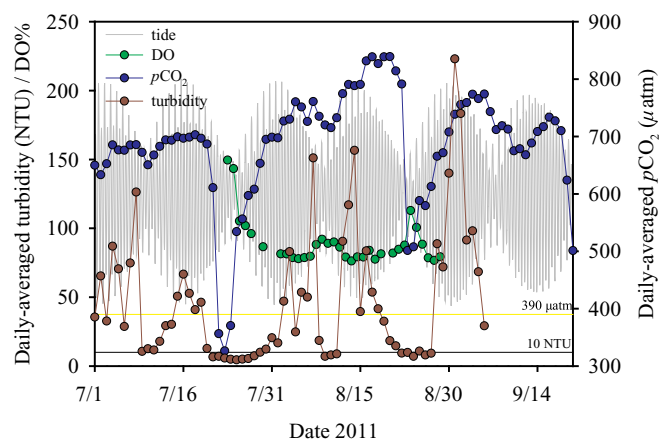
**Fig. 11.** Daily-averaged  $p\text{CO}_{2,\text{nontherm}}$  vs daily-averaged dissolved oxygen (DO) at the mooring site in winter (January 18 to February 17), spring (March 19 to May 14), and summer (July 23 to August 29), 2011. Black lines are the regression lines of these three seasons. The double arrow line denotes the  $p\text{CO}_2$  variation trend with DO following classic Redfield Stoichiometry.



**Fig. 12.** Variations in  $p\text{CO}_2$  ( $\delta p\text{CO}_2$ ) and term contributions from temperature, sea-air  $\text{CO}_2$  exchange, horizontal mixing, biological activity, and residual terms during April 3–30, August 14–18 (spring tide), and August 20–26 (neap tide), 2011.



**Fig. 13.** Turbidity versus chlorophyll *a* (Chl *a*) and  $p\text{CO}_2$  versus Chl *a* in surface seawater at the mooring site sampled on July 26, 2012.



**Fig. 14.** Tidal height variations and daily-averaged turbidity, dissolved oxygen (DO) saturation, and  $p\text{CO}_2$  at the mooring site sampled from July 1 to September 20, 2011. Yellow line is for atmospheric  $p\text{CO}_2$ . (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

system at the same site in July 2012, which indicated that when turbidity was  $< 10$  NTU, Chl *a* increased and  $p\text{CO}_2$  was reduced dramatically along with the decline in turbidity (Fig. 13). During the neap tides of summer 2011, turbidity was nearly 10 NTU or less, and sea surface  $p\text{CO}_2$  at the site exhibited a significant drawdown, at times even becoming lower than atmospheric  $p\text{CO}_2$  (Fig. 14). Meanwhile, DO saturation rate showed a large increase, at times even appearing to be supersaturated (Fig. 14). An algal bloom was observed near the buoy site during neap tide on August 22–24, 2011 (Liu, 2014), corresponding with a remarkable decrease in both  $p\text{CO}_2$  and turbidity. Generally, turbidity fluctuated with the tidal cycles, with high values during spring tides and low values during neap tides (Fig. 14). During spring tides, high turbidity resulted from strong current velocities and associated sediment resuspension. The mass balance model was separately applied to August 20–25 (neap tide in summer) when the algal bloom occurred, and to August 14–18 (spring tide in summer). During the spring tide in August, biological activity was the main driver of changes in surface water  $p\text{CO}_2$ , which resulted in  $p\text{CO}_2$  increases as high as  $123 \pm 49$   $\mu\text{atm}$  (Fig. 12), 60% of the total  $p\text{CO}_2$  change, suggesting respiration exceeded photosynthesis. The net community production (NCP) in the mixed layer was estimated to  $-17 \pm 10$   $\text{mmol m}^{-2} \text{d}^{-1}$ . The negative value suggests that this site was heterotrophic during August 14–18. The calculation method of NCP was presented in Appendix S1. Based on in situ incubation experiments conducted during the August 2011 cruise, the respiration rate was greater than primary productivity (Liu, 2014): i.e. the system was heterotrophic, supporting our model results. Similar results were also observed near our study area in June and August 2003 (Chen et al., 2006a,b). As opposed to other seasons, air-sea  $\text{CO}_2$  exchange significantly decreased  $p\text{CO}_2$  by  $68 \pm 9$   $\mu\text{atm}$  (Fig. 12), accounting for  $-37\%$  (degassing) of the total  $p\text{CO}_2$  change, since sea surface  $p\text{CO}_2$  was extremely high with an average value of  $815 \pm 23$   $\mu\text{atm}$  during spring tide in August. During the algal bloom event, which occurred at an August neap tide, we observed a dramatic decline in sea surface  $p\text{CO}_2$  (with an average of  $260 \pm 130$   $\mu\text{atm}$  for sea surface  $p\text{CO}_2$ ), which can largely be explained by the amplified primary production. Consistently, the model result showed that biological production was the greatest contributor ( $123 \pm 56$   $\mu\text{atm}$ ,  $-51\%$  of the absolute  $p\text{CO}_2$  change, Fig. 12) to the decrease in  $p\text{CO}_2$ . The mean NCP in the mixed layer during this period was  $18 \pm 9$   $\text{mmol m}^{-2} \text{d}^{-1}$ , much higher than that in April which was  $0.2 \pm 10$   $\text{mmol m}^{-2} \text{d}^{-1}$ . In addition,  $\text{CO}_2$  degassing reduced  $p\text{CO}_2$  by  $90 \pm 10$   $\mu\text{atm}$  (Fig. 12),  $-33\%$  of the absolute  $p\text{CO}_2$  change, suggesting that air-sea exchange also had an important effect on the  $\text{CO}_2$  decline. Overall, horizontal mixing was

a minor factor altering  $p\text{CO}_2$  in August. A weak relationship existed between temperature-normalized  $p\text{CO}_2$  and salinity in August ( $R^2 = 0.07$ , Fig. 9b), supporting our model result.

#### 4.4. Controls of the diurnal variations in surface seawater $p\text{CO}_2$

Temperature-induced diurnal variations in sea surface  $p\text{CO}_2$  ( $p\text{CO}_{2,\text{therm}}$ ) were smaller than those caused by non-temperature effects ( $p\text{CO}_{2,\text{nontherm}}$ ) during all seasons (Appendix Fig. S9), particularly in spring, suggesting that temperature had a limited effect on the diurnal variations in sea surface  $p\text{CO}_2$ .

In this study, sea surface  $p\text{CO}_2$  showed an inverse correlation with tidal height during winter, spring and spring-tides in summer, implying that tide-induced mixing between nearshore water (high  $p\text{CO}_2$ ) and offshore ECS water (low  $p\text{CO}_2$ ) likely dominate the diurnal variations in sea surface  $p\text{CO}_2$ . The diurnal amplitudes of sea surface  $p\text{CO}_2$  observed in winter and spring showed good correlations with SSS, with Pearson correlation coefficients of 0.88 and 0.89, respectively ( $n = 7$ ,  $P < .1$ ). This again suggests that tide-induced water mixing at our site played the most important role on the diurnal variations in surface  $p\text{CO}_2$ . Tidal mixing dominated the variations in sea surface  $p\text{CO}_2$  as has also been observed in Shenhui Bay and Xiamen Bay (Dai et al., 2009). It should be noted that diurnal amplitudes of  $p\text{CO}_2$  had insignificant or weaker correlations with diurnal tidal amplitudes in winter and spring, which might be explained by sporadic effects of the river plumes in addition to tidal-driven water mass mixing.

A typical biological cycle (photosynthesis and respiration) would lead to sea surface  $p\text{CO}_2$  drawdown during the day and increases at night, typically with mirrored variations in DO (Dai et al., 2009; Jiang et al., 2011). In the present study, neither biology-induced  $p\text{CO}_2$  variability nor significant inverse correlations between  $p\text{CO}_2$  and DO were observed in winter and spring, suggesting that photosynthesis and respiration may have little effect on the diurnal variations in sea surface  $p\text{CO}_2$  during these two seasons. However, there was one exception. During the neap tides of summer, diurnal variability in  $p\text{CO}_2$  was consistent with day-night cycles and mirrored DO variations.

#### 4.5. Controls on the inter-summer variation in surface water $p\text{CO}_2$

During the buoy mooring period, two full months of August were covered in 2010 and 2011. The corresponding monthly air-sea fluxes were  $11 \pm 13$  and  $34 \pm 69$   $\text{mmol C m}^{-2} \text{d}^{-1}$ , respectively. Extremely high variability ( $\pm 69$ ) was observed in August 2011, which was mainly due to high wind speeds on August 6 and 7 resulting from the influence of a typhoon. If we remove these two high wind speed days, the monthly air-sea flux estimate is reduced to  $21 \pm 17$   $\text{mmol C m}^{-2} \text{d}^{-1}$  in August 2011, which is still greater than that estimated for August 2010 according to a  $t$ -test ( $p < .05$ ). The higher  $\text{CO}_2$  exchange flux in August 2011 was caused by the high sea surface  $p\text{CO}_2$  values and wind speeds in 2011 relative to 2010 ( $727 \pm 110$  vs.  $619 \pm 135$   $\mu\text{atm}$ ), since comparable atmospheric  $p\text{CO}_2$  and sea surface temperatures were observed during these two months. Detailed monthly values are shown in Table 2.

Comparing surface  $p\text{CO}_2$  variations in August 2010 with August 2011 (Fig. 3e), we notice a drastic reduction in surface  $p\text{CO}_2$  occurring

on August 4–6, 20–21, 30–31, 2010, and August 23–24, 2011, all of which occurred during neap tides. August 2011 covered two neap tides, although one was affected by a typhoon, while August 2010 included ca. three neap tides, indicating stronger tidal currents existed in August 2011. If these low surface water  $p\text{CO}_2$  values during neap tide are excluded, the surface  $p\text{CO}_2$  was still significantly greater in August 2011 compared with August 2010, verified by  $t$ -test. The recalculated monthly average surface  $p\text{CO}_2$  values are  $696 \pm 76$  and  $763 \pm 71$   $\mu\text{atm}$  during August 2010 and 2011, respectively. In addition, if we assume SSS primarily represents riverine inputs, the lower SSS values (by 1.2 units) in August 2010 compared to August 2011 suggests higher river flows in August 2010. Consistently, the measured river discharge from the Changjiang at the Datong gauge station was greater in August 2010 than in August 2011. As shown in Section 4.3, TSM concentrations increase with strong tidal current velocities and lower river flows. TSM concentrations were retrieved from the obtained remote sensing reflectance based on the regional retrieval algorithm (He et al., 2013). Because of the heavy influence of clouds on the daily product, all daily products were merged into a monthly product as shown in Appendix Fig. S10. Here, we can see the TSM concentration was lower in August 2010 than in August 2011. Due to light limitation at the buoy site, it is mostly heterotrophic; lower TSM during neap tides would enhance biological uptake and drawdown surface  $p\text{CO}_2$ . Horizontal mixing was an insignificant factor altering sea surface  $p\text{CO}_2$  in summer (Chen et al., 2008). Here, we ruled out the influence from mixing between the high  $p\text{CO}_2$  Changjiang River and low  $p\text{CO}_2$  offshore seawater. Therefore, the relatively lower sea surface  $p\text{CO}_2$  in August 2010 was likely due to enhanced biological production, resulting from the lower turbidity driven by weak tidal currents and high river flows. Contrary results have been reported in the central South Atlantic Bight, which had higher aqueous  $p\text{CO}_2$  values and river fluxes from September–October 2007 compared with September–October 2006. This high surface  $p\text{CO}_2$  in fall 2007 is largely explained by enhanced decomposition of organic matter carried by saltmarshes and estuaries during increases in river flows (Xue et al., 2016). Moreover, in August 2010, under-sampling during neap tides would lead to a 12% increase in average surface  $p\text{CO}_2$  estimates and a 21% increase in estimated  $\text{CO}_2$  exchange fluxes. As a consequence, to avoid overestimating sea-to-air  $\text{CO}_2$  exchange fluxes in the Hangzhou Bay nearshore system, it is important to thoroughly characterize periods of low turbidity during neap tides.

## 5. Conclusions

The yearly continuous high frequency observations of sea surface  $p\text{CO}_2$  were conducted near the mouth of Hangzhou Bay (Outer Changjiang Estuary) from July 30 to October 20, 2010, and January 17 to September 20, 2011. Highly variable seawater  $p\text{CO}_2$  was observed over different timescales, ranging from seasonal to short-term scales. Over spring-neap and diel timescales, the mean amplitudes of seawater  $p\text{CO}_2$  were the lowest in winter, had medium values in spring, and the highest in summer. The mean amplitude of daily-averaged sea surface  $p\text{CO}_2$  during the summer was comparable to the  $p\text{CO}_2$  amplitude at the seasonal scale ( $283 \pm 139$  vs  $305 \pm 111$   $\mu\text{atm}$ ). In winter, this site

**Table 2**

Comparison of monthly surface water  $p\text{CO}_2$  and other parameters during August 2010 and August 2011 at the buoy site. Error bar represents one standard deviation.

Month	SSS	SST(°C)	Sea surface $p\text{CO}_2$ ( $\mu\text{atm}$ )	Atmospheric $p\text{CO}_2$ ( $\mu\text{atm}$ )	Wind velocity ( $\text{m s}^{-1}$ )	$\text{CO}_2$ flux <sup>a</sup> ( $\text{mmol m}^{-2} \text{d}^{-1}$ )	Changjiang flux <sup>b</sup> ( $\text{m}^3 \text{s}^{-1}$ )
August 2010	$22.8 \pm 2$	$26.6 \pm 1$	$619 \pm 135$	$380 \pm 9$	$4.1 \pm 0.4$	$11 \pm 13$	51,557
August 2011	$24.0 \pm 3$	$26.1 \pm 1$	$728 \pm 108$ ( $727 \pm 110$ )	$380 \pm 8$	$5.3 \pm 4$ ( $4.7 \pm 2$ )	$34 \pm 69$ ( $21 \pm 17$ )	30,288

Values in parenthesis denote monthly averages without two days (August 6 and 7) which were affected by an episodic event (typhoon).

<sup>a</sup> Gas transfer velocity of  $\text{CO}_2$  was calculated from Sweeney et al. (2007).

<sup>b</sup> The Changjiang flux was derived from the Datong gauge station (<http://xxfb.hydroinfo.gov.cn/>).

**Table 3**

Controlling factors on the variability of surface water  $p\text{CO}_2$  at different time scales.

Time scales	Controlling factors
Seasonal range	Temperature, water masses mixing
Intra-seasonal scales (spring-neap/ neap-spring tide)	
Winter	Temperature, water masses mixing
Spring	Temperature, water masses mixing, air-sea exchange
Summer	Biological metabolism, air-sea exchange
Diurnal range	
Winter	Tidal mixing
Spring	Tidal mixing
Summer (spring-tide)	Tidal mixing
Summer (neap-tide)	Biological uptake
Inter-summer	Biological metabolism associated with turbidity

was a weak atmospheric  $\text{CO}_2$  sink with an average air-sea exchange flux of  $-3.3 \pm 2 \text{ mmol C m}^{-2} \text{ d}^{-1}$ . However, in spring and summer, it shifted from a  $\text{CO}_2$  sink to a source, with averages air-sea  $\text{CO}_2$  fluxes of  $9.9 \pm 7$  and  $30 \pm 26 \text{ mmol C m}^{-2} \text{ d}^{-1}$ , respectively. Overall, the average sea surface  $p\text{CO}_2$  value was  $545 \pm 138 \mu\text{atm}$ , and the air-sea  $\text{CO}_2$  flux was estimated as  $14 \pm 9 \text{ mmol C m}^{-2} \text{ d}^{-1}$  during the observation period in 2011.

Driving factors that modulate seawater  $p\text{CO}_2$  in nearshore systems generally include temperature, air-sea exchange, mixing, and biological process. The controlling factors at different time scales were presented in Table 3. Over seasonal timescales, temperature and mixing were the dominant controls on  $p\text{CO}_2$  variability. Over spring-neap tidal cycle timescales, seawater  $p\text{CO}_2$  negatively correlated with SSS in winter and spring, demonstrating the significant contribution from water mass mixing during these two seasons. In addition, temperature was also an important factor affecting  $p\text{CO}_2$  in winter and spring. Based on mass balance models, temperature, mixing, and air-sea exchange were the major controlling factors, which generally supported our qualitative analysis. However, during the spring tides of August, biological metabolism and air-sea exchange were the dominant driving forces altering  $p\text{CO}_2$ . Over diurnal timescales, tide-induced mixing dominated the diurnal variations on surface  $p\text{CO}_2$  in winter, spring, and during spring tides of summer. However, during the neap tides of summer, biological metabolism played an important role in the diurnal variability of surface  $p\text{CO}_2$ . Microtidal and low turbidity estuarine systems generally have lower surface seawater  $p\text{CO}_2$  compared with macrotidal and highly turbid nearshore waters. For instance, the lower Neuse River estuary was found to be a  $\text{CO}_2$  sink with a flux of  $-1.37 \text{ mmol m}^{-2} \text{ d}^{-1}$  (Crosswell et al., 2012), while our study site was a source of atmospheric  $\text{CO}_2$ , with high turbidity limiting biological production.

The low turbidity values were found during the neap tides of summer, corresponding with weak current velocities and high river flows at our study site. Relative to August 2011, August 2010 had comparable SST and lower SSS values, but significantly lower sea surface  $p\text{CO}_2$  values. This lower  $p\text{CO}_2$  ( $\sim 100 \mu\text{atm}$ ) during August 2010 was likely caused by enhanced biological uptake due to weak currents and increased river flows.

This study highlights the significance of tidal mixing and tidal phases on the variability of surface water  $p\text{CO}_2$  from seasonal to diel scales. Even during the neap tides of summer, where biological activity dominates  $p\text{CO}_2$  changes, the tidal phase in essence reduces turbidity and therefore enhances biological uptake. Temporal under-sampling could thus impact the evaluation of air-sea  $\text{CO}_2$  fluxes. We contend that capturing the full tidal phase, particularly during periods of neap tide, is critically important for better understanding the processes and fluxes that govern air-sea  $\text{CO}_2$  exchanges in the many macrotidal nearshore systems of the world.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.marchem.2019.103690>.

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